

DESCRIPTION

Substituted Sym-triindole

5 Technical Field

[0001]

The present invention provides a novel substituted Sym-triindole derivative and a process for production thereof. The substituted Sym-triindole derivative of the present invention is applicable to a wide spectrum of uses, such as various electrification preventions, electrification controls, capacitors, batteries, chemical sensors, displays, organic EL materials, solar cells, photodiodes, phototransistors, nonlinear materials, photorefractive materials, rustproof agents, adhesives, fibers, antistatic paints, electrodeposition paints, plating primers, electric corrosion protections and the like.

Background Art

20 [0002]

Di-substituted Sym-triindole derivatives are reported. However, with the synthesis process therefor, it has been difficult to synthesize a tri- or higher-substituted Sym-triindole derivative. (See Patent Literature 1.)

5 [0003]

Patent Literature 1: JP-A-2000-105139

Disclosure of the Invention

Task to Be Achieved by the Invention

10 [0004]

The present invention aims at providing a novel substituted Sym-triindole derivative which is applicable to a wide spectrum of uses, such as various electrification preventions, electrification controls, capacitors, batteries, chemical
15 sensors, displays, organic EL materials, solar cells, photodiodes, phototransistors, nonlinear materials, photorefractive materials, rustproof agents, adhesives, fibers, antistatic paints, electrodeposition paints, plating primers, electric corrosion protections and the like.

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Means for Achieving the Task

[0005]

In view of the above-mentioned situation, the present inventor made a study on the process for producing a substituted Sym-triindole derivative. As a result, it was found unexpectedly that, by reacting a substituted oxyindole with a phosphorus oxyhalide such as phosphorus oxychloride or the like, a tri- or higher-substituted Sym-triindole derivative can be synthesized selectively, that is, the above-mentioned task can be achieved. The finding has led to the completion of the present invention.

Effect of the Invention

[0006]

By the process of the present invention, there can be provided a novel substituted Sym-triindole derivative which is applicable to a wide spectrum of uses, such as various electrification preventions, electrification controls, capacitors, batteries, chemical sensors, displays, organic EL materials, solar cells, photodiodes, phototransistors,

nonlinear materials, photorefractive materials, rustproof agents, adhesives, fibers, antistatic paints, electrodeposition paints, plating primers, electric corrosion protections and the like.

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Brief Description of the Drawings

[0007]

Fig. 1 shows an absorption spectrum in a visible-to-ultraviolet light region of Sym-N-(n-hexyl)-5-(4-formylphenyl)triindole (formula 36).

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Fig. 2 is a fluorescent spectrum of Sym-N-(n-hexyl)-5-(4-formylphenyl)triindole (formula 36) in a dichloromethane solution state.

Fig. 3 is a fluorescent spectrum of Sym-N-(n-hexyl)-5-(4-formylphenyl)triindole (formula 36) in a solid state.

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Best Mode for Carrying Out the Invention

[0008]

The present invention is described in detail below.

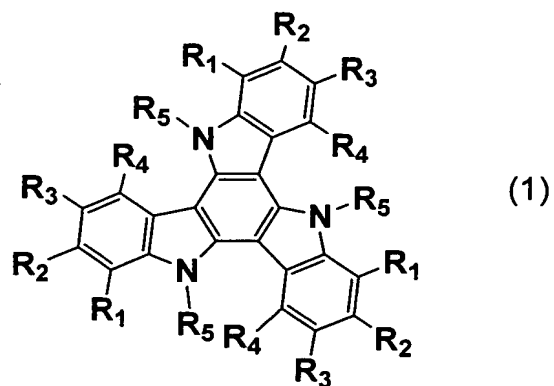
20 The present invention provides inventions described in the

following [1] to [14].

[0009] to [0011]

[1] A substituted Sym-triindole derivative represented by the following general formula (1)

5 [formula 1]



(wherein R₁, R₂, R₃ and R₄ are each independently hydrogen, halogen, C1-C6 alkyl group, C1-C6 haloalkyl group, substituted C1-C6 alkyl group, C2-C6 alkenyl group, substituted C2-C6 alkenyl group, C2-C6 alkynyl group, substituted C2-C6 alkynyl group, hydroxyl group, C1-C6 alkoxy group, aryloxy group, amino group, mono-substituted amino group, di-substituted amino group, acylamino group, mercapto group, C1-C6 alkylsulfenyl group, C1-C6 haloalkylsulfenyl group, arylsulfenyl group, substituted arylsulfenyl group, C1-C6 alkylsulfinyl group, C1-C6 haloalkylsulfinyl group, aralkylsulf-

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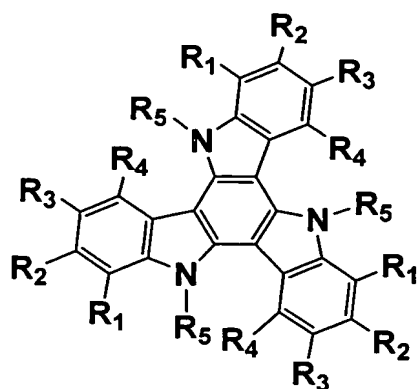
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enyl group, arylsulfinyl group, substituted arylsulfinyl group, C1-C6 alkylsulfonyl group, C1-C6 haloalkylsulfonyl group, arylsulfonyl group, substituted arylsulfonyl group, sulfonic acid group (hydroxysulfonyl group), aryl group, substituted aryl group, cyano group, nitro group, formyl group, acyl group, carboxyl group, C1-C6 alkoxy carbonyl group, carbamoyl group, N-mono-substituted carbamoyl group, N,N-disubstituted carbamoyl group, hydrazonomethyl group ($-\text{CH}=\text{N}-\text{NH}_2$ group), N-mono-substituted hydrazonomethyl group, N,N-disubstituted hydrazonomethyl group, oximemethyl group (hydroxyiminomethyl group), C1-C6 alkoxyiminomethyl group, or aryloxyiminomethyl group; R_5 is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group, or aryl C1-C6 alkyl group; in no event, all of R_1 , R_2 , R_3 and R_4 are hydrogen simultaneously).

[0012] to [0016]

[2] A process for producing a substituted Sym-triindole derivative represented by the following general formula (1)

[formula 3]

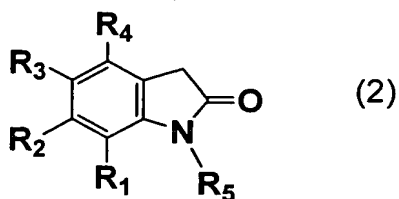


(1)

(wherein R₁, R₂, R₃ and R₄ are each independently hydrogen, halogen, C1-C6 alkyl group, C1-C6 haloalkyl group, substituted C1-C6 alkyl group, C2-C6 alkenyl group, substituted C2-C6 alkenyl group, C2-C6 alkynyl group, substituted C2-C6 alkynyl group, hydroxyl group, C1-C6 alkoxy group, aryloxy group, amino group, mono-substituted amino group, disubstituted amino group, acylamino group, mercapto group, C1-C6 alkylsulfenyl group, C1-C6 haloalkylsulfenyl group, aralkylsulfenyl group, arylsulfenyl group, substituted arylsulfenyl group, C1-C6 alkylsulfinyl group, C1-C6 haloalkylsulfinyl group, arylsulfinyl group, substituted arylsulfinyl group, C1-C6 alkylsulfonyl group, C1-C6 haloalkylsulfonyl group, arylsulfonyl group, substituted arylsulfonyl group, sulfonic acid group (hydroxysulfonyl group), aryl group, substituted aryl group, cyano group, nitro group, formyl group,

acyl group, carboxyl group, C1-C6 alkoxy carbonyl group, carbamoyl group, N-mono-substituted carbamoyl group, N,N-disubstituted carbamoyl group, hydrazonomethyl group ($-\text{CH}=\text{N}-\text{NH}_2$ group), N-mono-substituted hydrazonomethyl group, N,N-disubstituted hydrazonomethyl group, oximemethyl group (hydroxyiminomethyl group), C1-C6 alkoxyiminomethyl group, or aryloxyiminomethyl group; R_5 is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group, or aryl C1-C6 alkyl group; in no event, all of R_1 , R_2 , R_3 and R_4 are hydrogen simultaneously), which process comprises reacting a substituted oxyindole represented by the following general formula (2)

[formula 2]



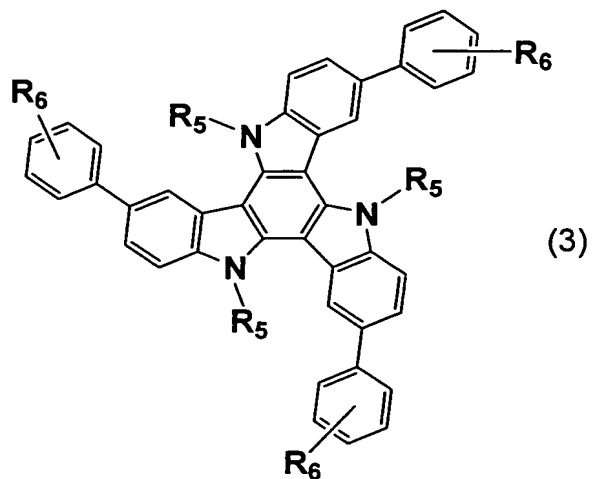
(wherein R_1 , R_2 , R_3 , R_4 and R_5 have the same definitions as given above) with a phosphorus oxyhalide.

[0017] to [0019]

[3] A Sym-triindole derivative represented by the following

general formula (3)

[formula 4]

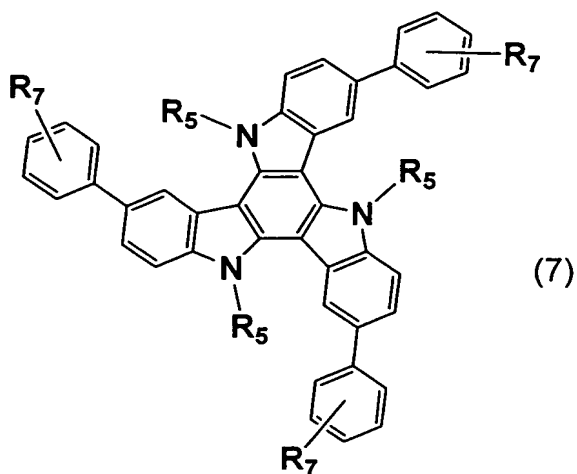


(wherein R_5 is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group, or aryl C1-C6 alkyl group; and
5 R_6 is hydrogen, formyl group, cyano group, C1-C6 alkoxy carbonyl group, dicyanovinyl group, aryl group or substituted aryl group).

[0020] to [0028]

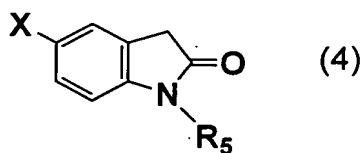
[4] A process for producing a Sym-triindole derivative represented by the following general formula (7)

[formula 8]



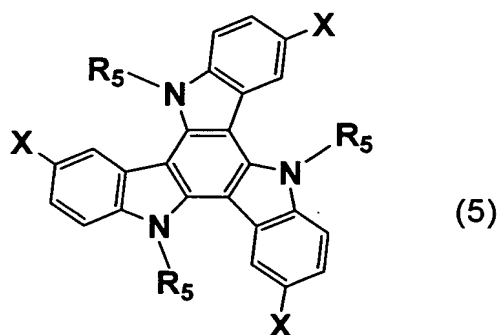
(wherein R_5 is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group; and R_7 is hydrogen, formyl group, cyano group, C1-C6 alkoxy carbonyl group, aryl group or substituted aryl group), which process comprises reacting an N-substituted-5-halo-oxyindole represented by the following general formula (4)

[formula 5]



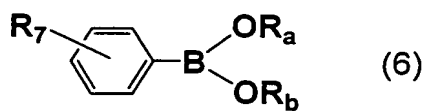
(wherein R_5 has the same definition as given above; and X is halogen) with a phosphorus oxyhalide to obtain an N-substituted-5-halo-triindole derivative represented by the following general formula (5)

[formula 6]



(wherein R₅ and X have the same definitions as given above) and further reacting it with a boric acid compound represented by the following general formula (6)

[formula 7]

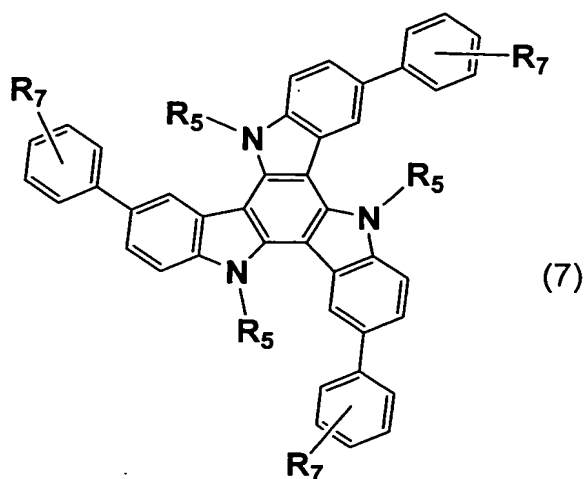


5 (wherein R₇ has the same definition as give above; and R_a and R_b are each independently hydrogen atom, C1-C6 alkyl group or optionally substituted phenyl group and may be combined to each other to form a ring).

[0029] to [0035]

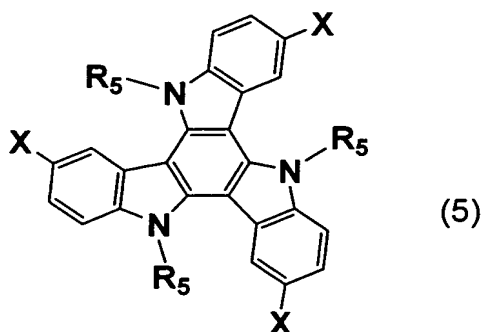
10 [5] A process for producing a Sym-triindole derivative represented by the following general formula (7)

[formula 11]



(wherein R_5 is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group; and R_7 is hydrogen, formyl group, cyano group, C1-C6 alkoxy-carbonyl group, aryl group or substituted aryl group), which
 5 process comprises reacting an N-substituted-5-halo-triindole derivative represented by the following general formula (5)

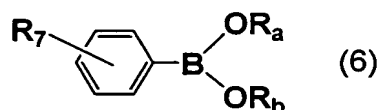
[formula 9]



(wherein R_5 has the same definition as given above; and X is halogen) with a boric acid compound represented by the fol-

lowing general formula (6)

[formula 10]

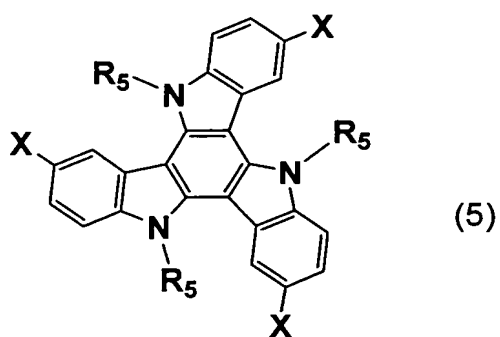


(wherein R_7 has the same definition as given above; and R_a and R_b are each independently hydrogen atom, C1-C6 alkyl group or optionally substituted phenyl group and may be combined to each other to form a ring).

[0036] to [0040]

[6] A process for producing an N-substituted-5-halo-triindole derivative represented by the following general
10 formula (5)

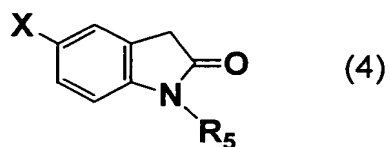
[formula 13]



(wherein R_5 is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group; and X is halogen), which process comprises reacting an N-

substituted-5-halo-oxyindole represented by the following
 general formula (4)

[formula 12]



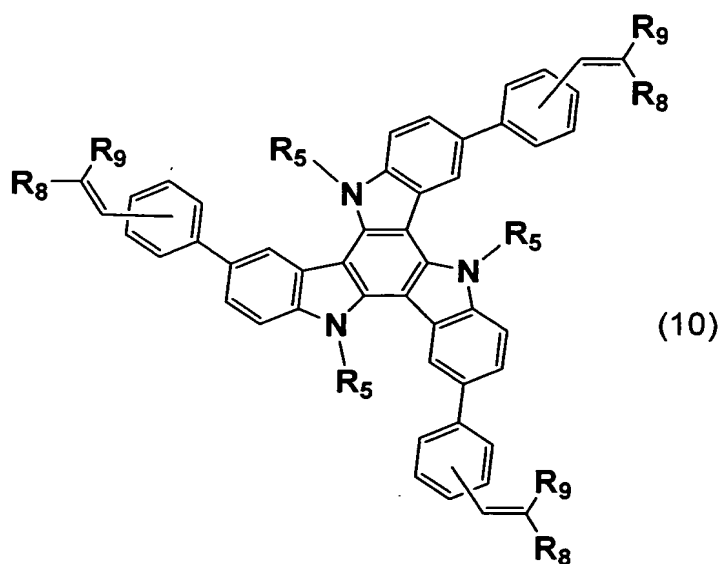
(wherein R₅ and X have the same definitions as given above)

5 with a phosphorus oxyhalide.

[0041] to [0047]

[7] A process for producing a Sym-triindole derivative represented by the following general formula (10)

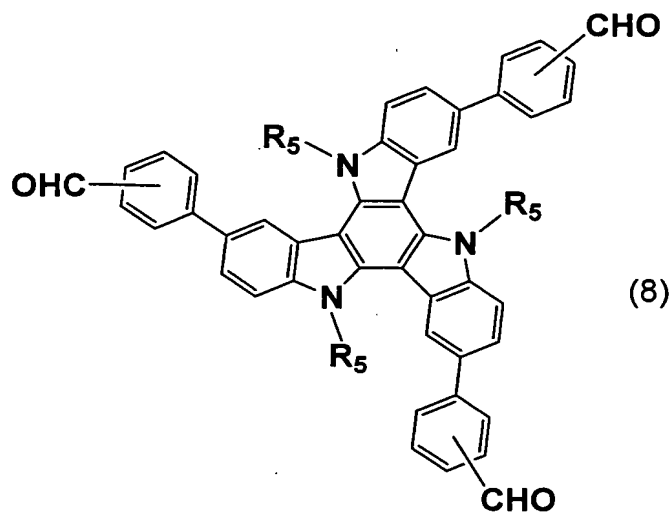
[formula 16]



10 (wherein R₅ is C2-C12 alkyl group, substituted C2-C12 alkyl

group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group; R₈ is hydrogen or cyano group; and R₉ is cyano group, carboxylic acid group, C1-C6 alkoxy carbonyl group, aryl group or substituted aryl group), which process comprises reacting a triindole derivative represented by the following general formula (8)

[formula 14]



(wherein R₅ has the same definition as given above) with a methylene compound represented by the general formula (9)

10 [formula 15]



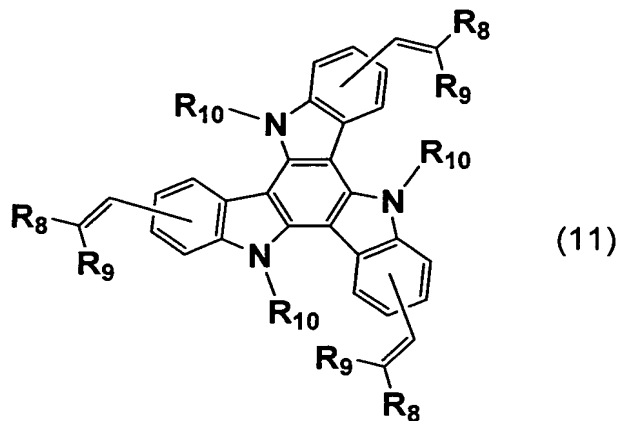
(wherein R₈ and R₉ have the same definitions as give above).

[0048] to [0050]

[8] A Sym-triindole vinyl derivative represented by the

following general formula (11)

[formula 17]

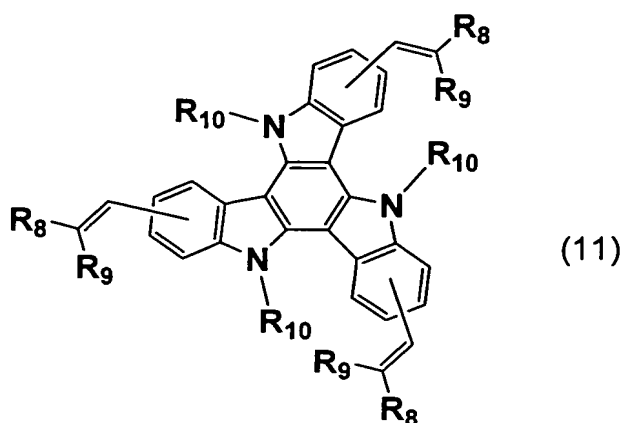


(wherein R_8 is hydrogen or cyano group; R_9 is cyano group, carboxylic acid group, C1-C6 alkoxy carbonyl group, aryl group
5 or substituted aryl group; and R_{10} is C₂-C₁₂ alkyl group, substituted C₂-C₁₂ alkyl group, C₂-C₁₂ haloalkyl group or aryl C1-C6 alkyl group).

[0051] to [0061]

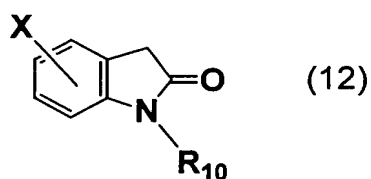
[9] A process for producing a Sym-triindole derivative represented by the following general formula (11)
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[formula 22]



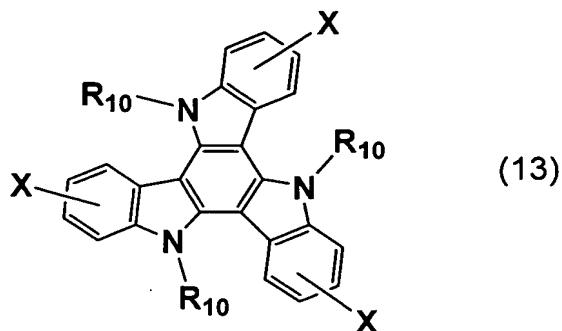
(wherein R₈ is hydrogen or cyano group; R₉ is cyano group, carboxylic acid group, C1-C6 alkoxy carbonyl group, aryl group or substituted aryl group; and R₁₀ is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group or
 5 aryl C1-C6 alkyl group), which process comprises reacting an oxyindole compound represented by the following general formula (12)

[formula 18]



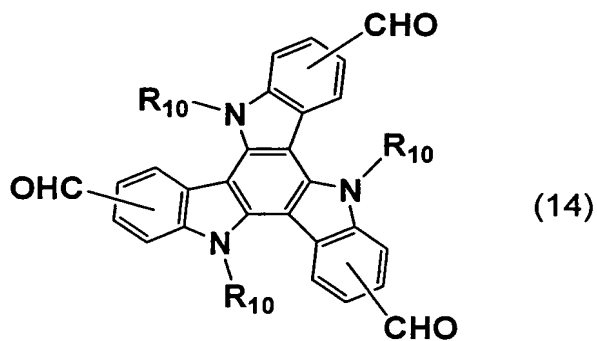
10 (wherein R₁₀ has the same definition as given above and X is halogen) with a phosphorus oxyhalide to obtain a Sym-halo-triindole derivative represented by the following general formula (13)

[formula 19]



(wherein R_{10} and X have the same definitions as given above),
 subjecting it to formylation with a formylating agent in the
 presence of butyllithium to obtain a Sym-formyltriindole de-
 5 rivative represented by the following general formula (14)

[formula 20]



(wherein R_{10} has the same definition as given above), and re-
 acting it with a methylene compound represented by the fol-
 lowing general formula (9)

10 [formula 21]



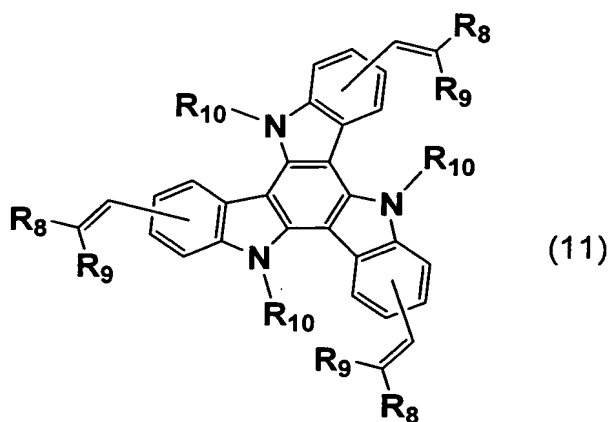
(wherein R₈ and R₉ have the same definitions as given above).

[0062] to [0068]

[10] A process for producing a Sym-triindole derivative represented by the following general formula (11)

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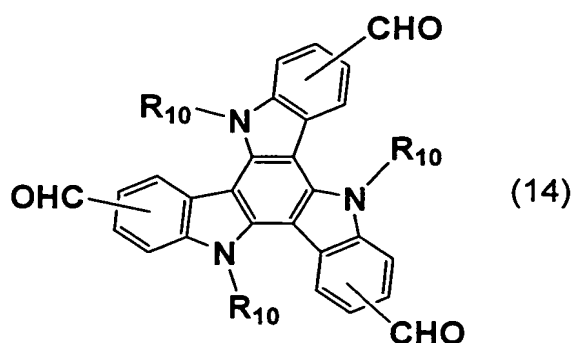
[formula 25]



(wherein R₈ is hydrogen or cyano group; R₉ is cyano group, carboxylic acid group, C1-C6 alkoxy carbonyl group, aryl group or substituted aryl group; and R₁₀ is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group or

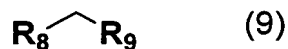
10 aryl C1-C6 alkyl group), which process comprises reacting a Sym-formyltriindole derivative represented by the following general formula (14)

[formula 23]



(wherein R_{10} has the same definition as given above) with a methylene compound represented by the following general formula (9)

[formula 24]

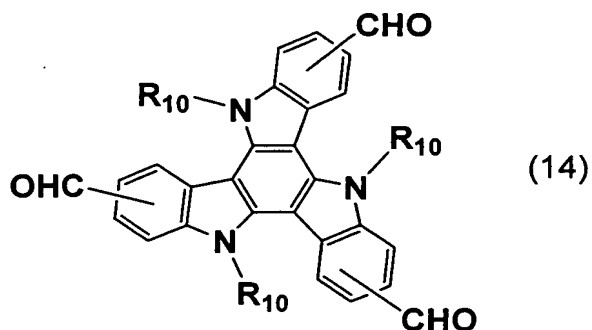


5 (wherein R_8 and R_9 have the same definitions as given above).

[0069] to [0073]

[11] A process for producing a Sym-formyltriindole derivative represented by the following general formula (14)

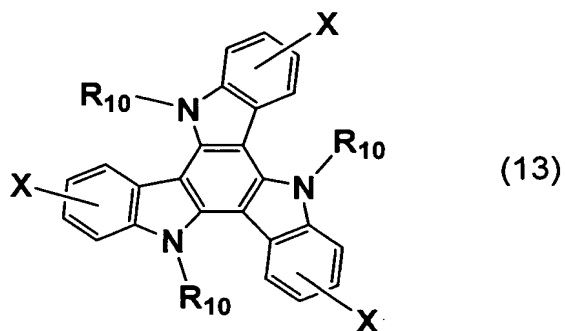
[formula 27]



10 (wherein R_{10} is C2-C12 alkyl group, C2-C12 substituted alkyl group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group),

which process comprises subjecting a Sym-halo-triindole derivative represented by the following general formula (13)

[formula 26]

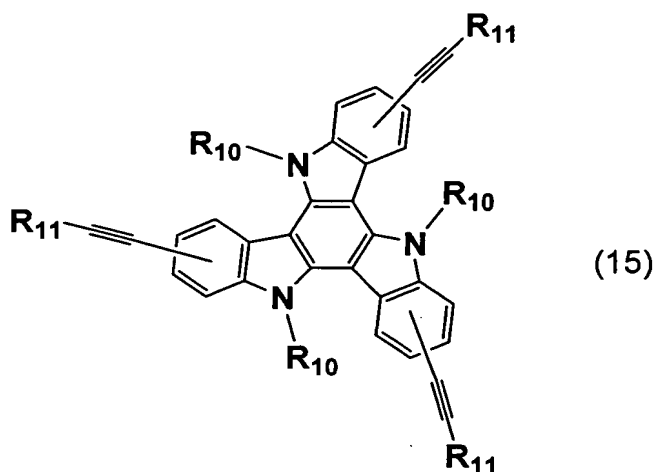


(wherein R_{10} has the same definition as given above and X is
 5 halogen), to formylation with a formylating agent in the
 presence of butyllithium.

[0074] to [0076]

[12] A Sym-triindole derivative represented by the following
 general formula (15)

10 [formula 28]

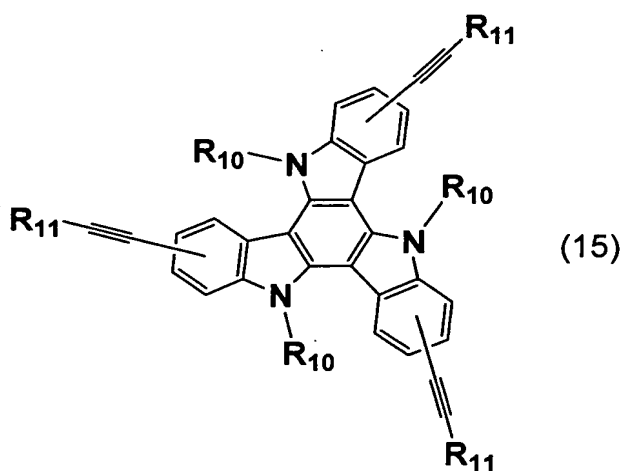


(wherein R_{10} is C2-C12 alkyl group, C2-C12 substituted alkyl group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group; and R_{11} is aryl group or substituted aryl group).

[0077] to [[0083]

- 5 [13] A process for producing a Sym-triindole derivative represented by the following general formula (15)

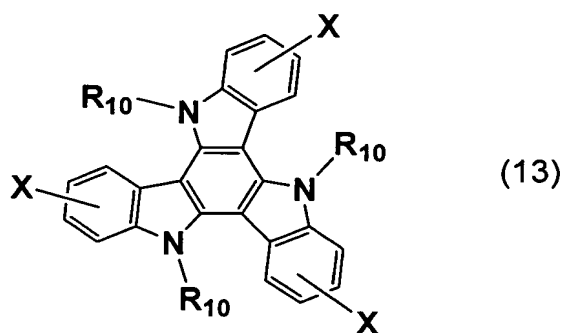
[formula 31]



(wherein R_{10} is C2-C12 alkyl group, substituted C2-C12 alkyl

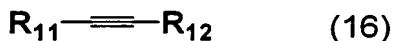
group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group and R₁₁ is aryl group or substituted aryl group), which process comprises reacting a Sym-halo-triindole derivative represented by the following general formula (13)

5 [formula 29]



(wherein R₁₀ has the same definition as given above and X is halogen) with an acetylene derivative represented by the following general formula (16)

[formula 30]

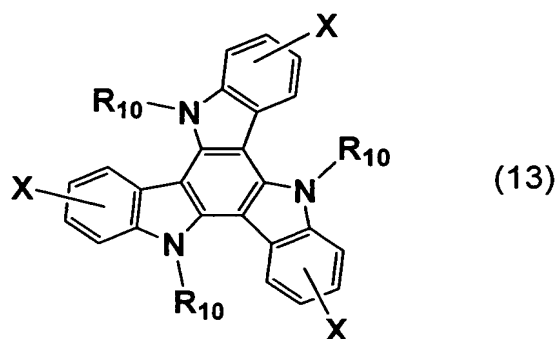


10 (wherein R₁₁ has the same definition as given above and R₁₂ is hydrogen or trimethylsilyl group).

[0084] to [0086]

[14] A Sym-halo-triindole derivative represented by the following general formula (13)

[formula 32]



(wherein R_{10} is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group or aryl C1-C6alkyl group; and X is halogen).

5 [0087]

Explanation is made on the terms used in the present specification.

[0088]

The expression of "C1-C6", etc. indicates that the
10 group following the expression has 1 to 6 carbon atoms, in
the case of "C1-C6".

[0089]

"Halogen" indicates fluorine, chlorine, bromine or io-
dine.

15 [0090]

"C1-C6 alkyl group" indicates straight chain or branched chain C1-C6 alkyl group having 1 to 6 carbon atoms and, as specific examples thereof, there can be mentioned methyl group, ethyl group, n-propyl group, isopropyl group, 5 n-butyl group, isobutyl group, sec-butyl group, tert-butyl group, n-pentyl group, n-hexyl group and 3,3-dimethylbutyl group.

[0091]

"C2-C12 alkyl group" indicates straight chain or 10 branched chain C2-C12 alkyl group having 2 to 12 carbon atoms and, as specific examples thereof, there can be mentioned methyl group, ethyl group, n-propyl group, isopropyl group, n-butyl group, isobutyl group, sec-butyl group, tert-butyl group, n-pentyl group, n-hexyl group, 3,3-dimethylbutyl group, 15 n-heptyl group, n-octyl group, 2-ethylhexyl group, n-decyl group and n-dodecyl group.

[0092]

"Aryl group" indicates aromatic group which may contain heterocyclic atom and which is a 5- or 6-membered single 20 ring or a condensed ring formed by any condensation of 5-

membered rings and/or 6-membered rings. There can be mentioned, for example, aromatic hydrocarbon groups of single ring or condensed ring, such as phenyl group, 1-naphthyl group, 2-naphthyl group, 1-anthranyl group, 2-anthranyl group, 5 9-anthranyl group, 1-phenanthryl group, 2-phenanthryl group, 9-phenanthryl group, 3-phenanthryl group, 4-phenanthryl, 1-pyrenyl group, 2-pyrenyl group, 4-pyrenyl group and the like; and aromatic heterocyclic ring group of single ring or condensed ring containing at least one nitrogen, oxygen or sulfur as heterocyclic atom, such as 2-pyridyl group, 3-pyridyl 10 group, pyrrole group, pyrazolyl group, triazolyl group, quinoliny group, indolyl group, isoindolyl group, thienyl group, benzothienyl group, furyl group, benzofuryl group, thiazolyl group, oxadiazolyl group, benzothiazolyl group, oxazolyl 15 group, isoxazolyl group, benzoxazolyl group, thiadiazolyl group, benzothiadiazolyl group and the like.

[0093]

"Substituted aryl group" indicates aryl group having the above-shown meaning, substituted independently with 1 to 20 7 groups selected from halogen having the above-mentioned

meaning, cyano group, formyl group and aryl group. There can be mentioned, for example, 2-chlorophenyl group, 4-cyanophenyl group, 4-formylphenyl group, 2-phenyl-1,3,4-oxadiazol-5-yl group, 2-(4-n-butylphenyl)-1,3,4-oxadiazol-5-yl group, 2-(4-tert-butylphenyl)-1,3,4-oxadiazol-5-yl group, and 2-(4-n-hexylphenyl)-1,3,4-oxadiazol-5-yl group.

[0094]

"C1-C6 alkoxy group" indicates "(C1-C6 alkyl having the above-shown meaning)-O- group". There can be mentioned, for example, methoxy group, ethoxy group, n-propoxy group, isopropoxy group, n-butoxy group, tert-butoxy group, n-pentyloxy group and n-hexyloxy group.

[0095]

"C1-C6 alkoxycarbonyl group" indicates "(C1-C6 alkoxy having the above-shown meaning)-CO- group". There can be mentioned, for example, methoxycarbonyl group and ethoxycarbonyl group.

[0096]

"Aryloxy group" indicates "(aryl having the above-shown meaning)-O- group". There can be mentioned, for example,

phenoxy group.

[0097]

“C1-C6 haloalkyl group” indicates straight chain or branched chain C1-C6 alkyl group substituted with 1 to 13
5 same or different halogens having the above-shown meaning. There can be mentioned, for example, chloromethyl group, dichloromethyl group, trichloromethyl group and trifluoromethyl group.

[0098]

10 “C2-C12 haloalkyl group” indicates straight chain or branched chain C2-C12 alkyl group substituted with 1 to 13 same or different halogens having the above-shown meaning. There can be mentioned, for example, chloroethyl group, 1,1-dichloroethyl group, 2,2,2-trichloroethyl group and 2,2,2-
15 trifluoroethyl group.

[0099]

“Substituted C1-C6 alkyl group” indicates straight chain or branched chain C1-C6 alkyl group substituted with 1 to 13 same or different hydroxy groups, C1-C6 alkoxy groups
20 having the above-shown meaning and aryl groups having the

above-shown meaning. There can be mentioned, for example, hydroxymethyl group, methoxymethyl group and phenylethyl group.

[0100]

5 "Substituted C2-C12 alkyl group" indicates straight chain or branched chain C2-C12 alkyl group substituted with 1 to 13 same or different hydroxy groups having the above-shown meaning and C1-C6 alkoxy groups having the above-shown meaning. There can be mentioned, for example, hydroxyethyl group
10 and methoxyethyl group.

[0101]

"Aryl C1-C6 alkyl group" indicates straight chain or branched chain C2-C12 alkyl group substituted with 1 to 13 same or different aryl groups having the above-shown meaning.
15 There can be mentioned, for example, benzyl group and phenylethyl group.

[0102]

"C2-C6 alkenyl group" indicates straight chain or branched chain C2-C6 alkenyl group. There can be mentioned,
20 for example, vinyl group, 1-propenyl group and 1-butenyl

group.

[0103]

"Carboxyl group" indicates carboxyl group or carboxyl group of salt form. There can be mentioned, for example, groups such as carboxylic acid (-COOH), sodium carboxylate (-COONa), potassium carboxylate (-COOK), lithium carboxylate (-COOLi), pyridinium carboxylate (-COOH·pyridine) and the like.

[0104]

"Substituted C2-C6 alkenyl group" indicates straight chain or branched chain C2-C6 alkenyl group substituted with 1 to 11 same or different cyano groups having the above-shown meaning, carboxylic acid groups having the above-shown meaning, C1-C6 alkoxycarbonyl groups having the above-shown meaning, aryl groups having the above-shown meaning, and substituted aryl groups having the above-shown meaning. There can be mentioned, for example, 2,2-dicyanovinyl group, (2-cyano-2-ethoxycarbonyl)vinyl group, (2-cyano-2-carboxy)vinyl group, (2-cyano-2-phenyl)vinyl group, (2-cyano-2-(4-nitrophenyl))vinyl group, and (2-cyano-2-(4-pyridyl))vinyl

group.

[0105]

"C2-C6 alkynyl group" indicates straight chain or branched chain C2-C6 alkynyl group. There can be mentioned, 5 for example, ethynyl group, 1-propynyl group and 1-butylnyl.

[0106]

"Substituted C2-C6 alkynyl group" indicates straight chain or branched chain C2-C6 alkynyl group substituted with 1 to 9 aryl groups having the above-shown meaning. There can 10 be mentioned, for example, phenylethynyl group and (4-pyridyl)ethynyl group.

[0107]

"Acyl group" indicates "(C1-C6 alkyl having the above-shown meaning)-CO- group", "C1-C6 haloalkyl group having the 15 above-shown meaning)-CO group, "aryl group having the above-shown meaning)-CO- group", or "substituted aryl group having the above-shown meaning)-CO- group". There can be mentioned, for example, acetyl group, propionyl group and benzoyl group.

[0108]

20 "Mono-substituted amino group" indicates amino group

mono-substituted with a C1-C6 alkyl group having the above-shown meaning or with an aryl group having the above-shown meaning. There can be mentioned, for example, methylamino group, ethylamino group and phenylamino group.

5 [0109]

"Di-substituted amino group" indicates amino group di-substituted with same or different C1-C6 alkyl groups having the above-shown meaning or same or different aryl groups having the above-shown meaning. There can be mentioned, for example, dimethylamino group, diethylamino group and diphenylamino group.

[0110]

"Acylamino group" indicates amino group mono-substituted with a C1-C6 acyl group having the above-shown meaning. There can be mentioned, for example, acetylamino group, propionylamino group and butyrylamino group.

[0111]

"C1-C6 alkylsulfenyl group" indicates "(C1-C6 alkyl having the above-shown meaning)-S- group". There can be mentioned, for example, methylsulfenyl group (methylthio group)

and ethylsulfenyl group (ethylthio group).

[0112]

"Arylsulfenyl group" indicates "(aryl having the above-shown meaning)-S- group". There can be mentioned, for example, phenylsulfenyl group (phenylthio group).

[0113]

"Substituted arylsulfenyl group" indicates "(substituted aryl group having the above-shown meaning)-S- group". There can be mentioned, for example, (4-chlorophenyl)sulfenyl group ((4-chlorophenyl)thio group).

[0114]

"C1-C6 haloalkylsulfenyl group" indicates "(C1-C6 haloalkyl having the above-shown meaning)-S- group". There can be mentioned, for example, trifluoromethylsulfenyl group (trifluoromethylthio group).

[0115]

"Aralkylsulfenyl group" indicates "(aryl having the above-shown meaning)-(alkyl having the above-shown meaning)-S- group" or "(substituted aryl having the above-shown meaning)-(alkyl having the above-shown meaning)-S- group". There

can be mentioned, for example, benzylsulfinyl group (benzylthio group), (4-chlorobenzyl)sulfinyl group ((4-chlorobenzyl)thio group).

[0116]

5 "C1-C6 alkylsulfinyl group" indicates "(C1-C6 alkyl having the above-shown meaning)-SO- group". There can be mentioned, for example, methylsulfinyl group and ethylsulfinyl group.

[0117]

10 "C1-C6 haloalkylsulfinyl group" indicates "(C1-C6 haloalkyl having the above-shown meaning)-SO- group". There can be mentioned, for example, trifluoromethylsulfinyl group, chloromethylsulfinyl group and 2-chloroethylsulfinyl group.

[0118]

15 "Arylsulfinyl group" indicates "(aryl having the above-shown meaning)-SO- group". There can be mentioned, for example, phenylsulfinyl group.

[0119]

"Substituted arylsulfinyl group" indicates "(substituted aryl having the above-shown meaning)-SO- group". There

can be mentioned, for example, 4-chlorophenylsulfinyl group.

[0120]

"C1-C6 alkylsulfonyl group" indicates "(C1-C6 alkyl having the above-shown meaning)-SO₂- group". There can be
5 mentioned, for example, methanesulfonyl group and ethanesulfonyl group.

[0121]

"C1-C6 haloalkylsulfonyl group" indicates "(C1-C6 haloalkyl having the above-shown meaning)-SO₂- group". There
10 can be mentioned, for example, trifluoromethanesulfonyl group, chloromethanesulfonyl group and 2-chloroethanesulfonyl group.

[0122]

"Arylsulfonyl group" indicates "(aryl having the above-shown meaning)-SO₂- group". There can be mentioned, for ex-
15 ample, phenylsulfonyl group.

[0123]

"Substituted arylsulfonyl group" indicates "(substituted aryl having the above-shown meaning)-SO₂- group".
There can be mentioned, for example, (4-chlorophenyl)sulfonyl
20 group.

[0124]

"Sulfonic acid group" indicates $\text{-SO}_2\text{-OH}$ group.

[0125]

"N-mono-substituted carbamoyl group" indicates carbamoyl group whose nitrogen atom is mono-substituted with a C1-C6 alkyl group having the above-shown meaning or with an aryl group having the above-shown meaning. There can be mentioned, for example, methylcarbamoyl group and phenylcarbamoyl group.

10 [0126]

"N,N-di-substituted carbamoyl group" indicates carbamoyl group whose nitrogen atom is di-substituted with C1-C6 alkyl groups having the above-shown meaning or with aryl groups having the above-shown meaning. There can be mentioned, for example, dimethylcarbamoyl group and diphenylcarbamoyl group.

[0127]

"Hydrazonomethyl group" indicates -CH=N-NH_2 group.

[0128]

20 "N-mono-substituted hydrazonomethyl group" indicates

hydrazonomethyl group whose nitrogen atom is mono-substituted with a C1-C6 alkyl group having the above-shown meaning or with an aryl group having the above-shown meaning. There can be mentioned, for example, methylhydrazonomethyl group and
5 phenylhydrazonomethyl group.

[0129]

"N,N-di-substituted hydrazonomethyl group" indicates hydrazonomethyl group whose nitrogen atom is di-substituted with C1-C6 alkyl groups having the above-shown meaning or
10 with aryl groups having the above-shown meaning. There can be mentioned, for example, dimethylhydrazonomethyl group and diphenylhydrazonomethyl group.

[0130]

"Oximemethyl group (hydroxyiminomethyl group)" indicates -CH=N-OH group. "C1-C6 alkoxyiminomethyl group" indicates oxyiminomethyl group whose oxygen atom is substituted with C1-C6 alkyl group having the above-shown meaning, and there can be mentioned, for example, methoxyiminomethyl group and ethoxyiminomethyl group.

20 [0131]

"Aryloxyiminomethyl group" indicates oxyiminomethyl group whose oxygen atom is substituted with aryl group having the above-shown meaning. There can be mentioned, for example, phenoxyiminomethyl group.

5 [0132]

The present invention is described in detail below.

[0133]

The present invention [1] lies in a Sym-triindole derivative represented by the general formula (1). In the general formula (1), R₁, R₂, R₃ and R₄ may be the same or different from each other and are each independently hydrogen, halogen, C₁-C₆ alkyl group, C₁-C₆ haloalkyl group, substituted C₁-C₆ alkyl group, C₂-C₆ alkenyl group, substituted C₂-C₆ alkenyl group, C₂-C₆ alkynyl group, substituted C₂-C₆ alkynyl group, hydroxyl group, C₁-C₆ alkoxy group, aryloxy group, amino group, mono-substituted amino group, di-substituted amino group, acylamino group, mercapto group, C₁-C₆ alkylsulfenyl group, C₁-C₆ haloalkylsulfenyl group, arylsulfenyl group, substituted arylsulfenyl group, C₁-C₆ alkylsulfinyl group, C₁-C₆ haloalkylsulfinyl group, aralkylsulfenyl group,

10
15
20

arylsulfinyl group, substituted arylsulfinyl group, C₁-C₆ alkylsulfonyl group, C₁-C₆ haloalkylsulfonyl group, arylsulfonyl group, substituted arylsulfonyl group, sulfonic acid group (hydroxysulfonyl group), aryl group, substituted aryl group, cyano group, nitro group, formyl group, acyl group, carboxyl group, C₁-C₆ alkoxy carbonyl group, carbamoyl group, N-mono-substituted carbamoyl group, N,N-di-substituted carbamoyl group, hydrazonomethyl group (-CH=N-NH₂ group), N-mono-substituted hydrazonomethyl group, N,N-di-substituted hydrazonomethyl group, oximemethyl group (hydroxyiminomethyl group), C₁-C₆ alkoxyiminomethyl group, or aryloxyiminomethyl group; R₅ is C₂-C₁₂ alkyl group, substituted C₂-C₁₂ alkyl group, C₂-C₁₂ haloalkyl group, or aryl C₁-C₆ alkyl group; in no event, all of R₁, R₂, R₃ and R₄ are hydrogen simultaneously.

[0134]

Therefore, as examples of the compound of the present invention [1], there can be mentioned compounds shown in Table 1. Incidentally, the abbreviations used in each of the following Tables have the following meanings.

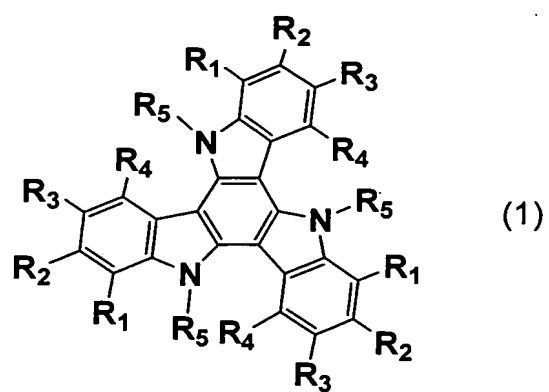
[0135]

	MeO: methoxy group
	COOMe: methoxycarbonyl group
	COOEt: ethoxycarbonyl
	Ph: phenyl group
5	NHBn: benzylamino group
	SPh: phenylsulfenyl group (phenylthio group)
	SOPh: phenylsulfinyl group
	SO ₂ Ph: phenylsulfonyl group
	COMe: acetyl group
10	CONHPh: phenylcarbamoyl group
	NHCOPh: benzoylamino group
	vinyl: vinyl group
	Et: ethyl group
	n-Pr: n-propyl group
15	Bu: n-butyl group
	Pen: n-pentyl group
	n-Hex: n-hexyl group
	(2-Et)Hex: 2-ethylhexyl group
	dicyanovinyl: 2,2-dicyanovinyl group
20	Py: pyridyl group

TMS-Ethynyl: trimethylsilylethynyl group

[0136]

[Formula 33]



[0137]

Table 1

Compound No.	R ₁	R ₂	R ₃	R ₄	R ₅	Melting point (°C)
1	Br	H	H	H	Et	205—207 161 160 185—187 243
2	H	Br	H	H	n-Pr	
3	H	H	Br	H	Bu	
4	H	H	H	Br	Penthyl	
5	H	H	Br	H	n-Hex	
6	H	H	Br	H	(2-Et)Hex	
7	H	H	Me	H	n-Hex	
8	H	H	CHO	H	n-Hex	
9	H	H	CHO	H	(2-Et)Hex	
10	H	H	CN	H	n-Hex	
11	H	H	COOMe	H	n-Hex	
12	H	COOH	H	H	n-Hex	
13	H	MeO	H	H	n-Hex	
14	H	H	NH ₂	H	n-Hex	
15	H	H	NHBn	H	n-Hex	
16	H	H	SPh	H	n-Hex	
17	H	H	SOPh	H	n-Hex	
18	H	H	SO ₂ Ph	H	n-Hex	
19	H	H	SO ₃ H	H	n-Hex	
20	H	H	COMe	H	n-Hex	
21	H	H	CONH ₂	H	n-Hex	
22	H	H	CONHPh	H	n-Hex	
23	H	H	NHCOPh	H	n-Hex	
24	H	H	NO ₂	H	n-Hex	
25	H	H	CF ₃	H	n-Hex	
26	H	H	Br	H	CF ₃ CH ₂	
27	H	H	TMS-Ethynyl	H	n-Hex	
28	H	H	Vinyl	H	n-Hex	
29	H	H	CN	H	(2-Et)Hex	

[0138]

The present invention [2] lies in a process for producing a substituted Sym-triindole derivative represented by the

general formula (1), which comprises reacting an oxyindole represented by the general formula (2) with a phosphorus oxyhalide.

[0139]

5 Firstly, description is made on the oxyindole represented by the general formula (2), used as a raw material of the present invention.

[0140]

In the general formula (2), R_1 , R_2 , R_3 and R_4 may be the
10 same or different from each other and are each independently hydrogen, halogen, C_1 - C_6 alkyl group, C_1 - C_6 haloalkyl group, substituted C_1 - C_6 alkyl group, C_2 - C_6 alkenyl group, substituted C_2 - C_6 alkenyl group, C_2 - C_6 alkynyl group, substituted C_2 - C_6 alkynyl group, hydroxyl group, C_1 - C_6 alkoxy group, ary-
15 loxy group, amino group, mono-substituted amino group, disubstituted amino group, acylamino group, mercapto group, C_1 - C_6 alkylsulfenyl group, C_1 - C_6 haloalkylsulfenyl group, aralkylsulfenyl group, arylsulfenyl group, substituted arylsulfenyl group, C_1 - C_6 alkylsulfinyl group, C_1 - C_6 haloalkyl-
20 sulfinyl group, arylsulfinyl group, substituted arylsulfinyl

group, C₁-C₆ alkylsulfonyl group, C₁-C₆ haloalkylsulfonyl group, arylsulfonyl group, substituted arylsulfonyl group, sulfonic acid group (hydroxysulfonyl group), aryl group, substituted aryl group, cyano group, nitro group, formyl group, acyl group, carboxyl group, C₁-C₆ alkoxy carbonyl group, carbamoyl group, N-mono-substituted carbamoyl group, N,N-disubstituted carbamoyl group, hydrazonomethyl group (-CH=N-NH₂ group), N-mono-substituted hydrazonomethyl group, N,N-disubstituted hydrazonomethyl group, oximemethyl group (hydroxyiminomethyl group), C₁-C₆ alkoxyiminomethyl group, or aryloxyiminomethyl group; R₅ is C₂-C₁₂ alkyl group, substituted C₂-C₁₂ alkyl group, C₂-C₁₂ haloalkyl group, or aryl C₁-C₆ alkyl group; in no event, all of R₁, R₂, R₃ and R₄ are hydrogen simultaneously.

[0141]

In the general formula (2), bromine is preferred as each halogen as R₁ to R₄, from the standpoints of reactivity and commercial availability; and C₂-12 alkyl group, particularly n-hexyl group or 2-ethylhexyl group is preferred as R₅.

[0142]

Therefore, there can be mentioned, as specific examples of the oxyindole usable in the above reaction, N-ethyl-5-bromooxyindole, N-ethyl-6-bromooxyindole, N-ethyl-7-bromooxyindole, N-ethyl-5,6-dibromooxyindole, N-ethyl-5,7-dibromooxyindole, N-ethyl-5-chlorooxyindole, N-ethyl-5-iodooxyindole, N-propyl-5-bromooxyindole, N-isopropyl-5-bromooxyindole, N-(n-hexyl)-5-bromooxyindole, N-(n-octyl)-5-bromooxyindole, N-(2-ethyl)hexyl-5-bromooxyindole, and N-(n-hexyl)-5-methyloxyindole.

10 [0143]

The oxyindole used in the present invention is known, or can be produced by reacting a corresponding aniline with chloroacetyl chloride in the presence of aluminum chloride.

[0144]

15 As the phosphorus oxyhalide, there can be mentioned, for example, phosphorus oxychloride and phosphorus oxybromide. Ordinarily, phosphorus oxychloride is used preferably.

[0145]

The molar ratio of the oxyindole used as a raw material
20 and the phosphorus oxyhalide may be any molar ratio as long

as stirring is possible. However, the phosphorus oxyhalide is used in an amount of, for example, preferably 0.5 to 30 liters, more preferably 1 to 10 liters relative to 1 mole of the oxyindole as raw material.

5 [0146]

The temperature of the reaction is not particularly restricted as long as the reaction proceeds at the temperature, but is preferred to be, for example, a range from 80°C to the reflux temperature.

10 [0147]

The period of the reaction is not particularly restricted but is preferred to be, for example, 1 to 48 hours.

[0148]

Incidentally, a compound wherein R_1 , R_2 , R_3 and R_4 are
15 hydrazonomethyl group, N-mono-substituted hydrazonomethyl group or N,N-di-substituted hydrazonomethyl group, can be produced by reacting a compound (obtained as above) wherein R_1 , R_2 , R_3 and R_4 are each formyl group, with hydrazine, an N-mono-substituted hydrazine [typified by a C1-C6 alkylhydra-
20 zine (e.g. methylhydrazine or ethylhydrazine) or phenylhydra-

zine], or an N,N-di-substituted hydrazine [typified by a di(C1-C6 alkyl)hydrazine (e.g. N,N-dimethylhydrazine or N,N-diethylhydrazine) or diphenylhydrazine].

[0149]

5 Also, a compound wherein R₁, R₂, R₃ and R₄ are oxime-methyl group (hydroxyiminomethyl group) or O-substituted oxymethyl group (e.g. C1-C6 alkoxyiminomethyl group or aryloxyiminomethyl group), can be produced by reacting a compound (obtained as above) wherein R₁, R₂, R₃ and R₄ are each
10 formyl group, with hydroxylamine or a salt thereof, or by reacting a compound (obtained as above) wherein R₁, R₂, R₃ and R₄ are each formyl group, with an O-substituted oxyamine such as C1-C6 alkoxyamine (alkanolamine) [e.g. methoxyamine (methanolamine) or salt thereof, or ethoxyamine (ethanolamine) of salt thereof], aryloxyamine (typified by
15 phenoxyamine), or the like.

[0150]

 The present invention [3] lies in a Sym-triindole derivative represented by the general formula (3). In the general formula (4), R₆ is hydrogen atom; formyl group; cyano
20

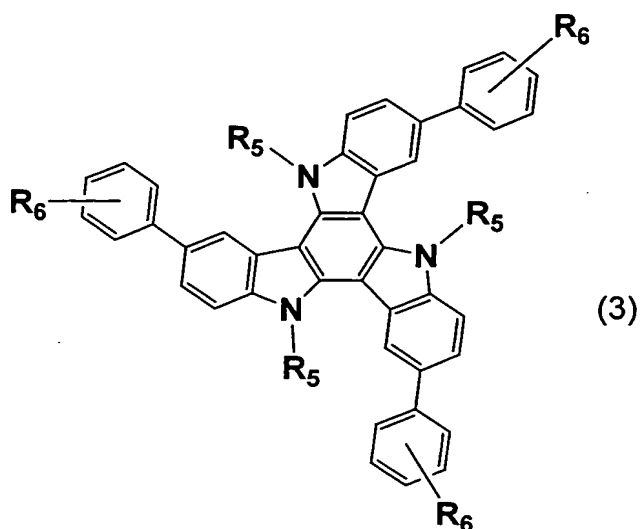
group; dicyanovinyl group; C1-C6 alkoxy carbonyl group such as methoxy carbonyl group, ethoxy carbonyl group, propoxy carbonyl group or the like; aryl group such as oxadiazolyl group or the like; or substituted aryl group such as 2-chlorophenyl group, 4-cyanophenyl group, 4-formylphenyl group, 2-phenyl-
5 1,3,4-oxadiazol-5-yl group, 2-(4-n-butylphenyl)-1,3,4-oxadiazol-5-yl group, 2-(4-tert-butylphenyl)-1,3,4-oxadiazol-5-yl group, 2-(4-n-hexylphenyl)-1,3,4-oxadiazol-5-yl group or the like.

10 [0151]

Therefore, compounds shown in Table 2 can be mentioned as examples of the compound of the present invention.

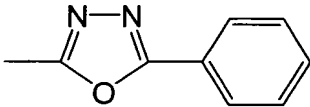
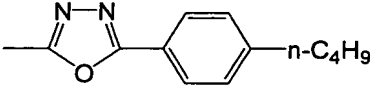
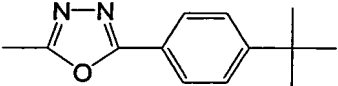
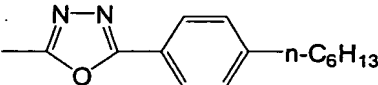
[0152]

[Formula 34]



[0153]

Table 2

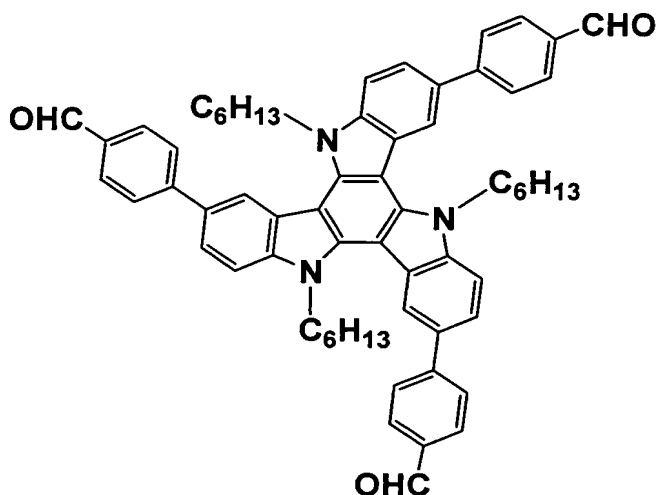
Compound No.	R ₅ (each of three R ₅ s)	R ₆ (each of three R ₆ s)	Melting point(°C)
30	n-Hex	H	245—257 203—213 297—303 281—291
31	n-Hex	4-CHO	
32	n-Hex	4-COOEt	
33	n-Hex	4-CN	
34	n-Hex	4-dicyanovinyl	
35	n-Hex	2-CHO	
36	n-Hex	3-CHO	
37	n-Hex	2-CN	
38	n-Hex	3-CN	
39	(2-Et)Hex		
40	(2-Et)Hex		
41	(2-Et)Hex		
42	(2-Et)Hex		

[0154]

As to Sym-N-(n-hexyl)-5-(4-formylphenyl)triindole (formula 35) which is a representative compound of the present invention, there are shown its absorption spectrum in a visible to ultraviolet region, in Fig. 1, and its fluorescent spectra in Fig. 2 (as measured in a dichloromethane solution) and Fig. 3 (as measured in a solid state).

[0155]

[Formula 35]



10 [0156]

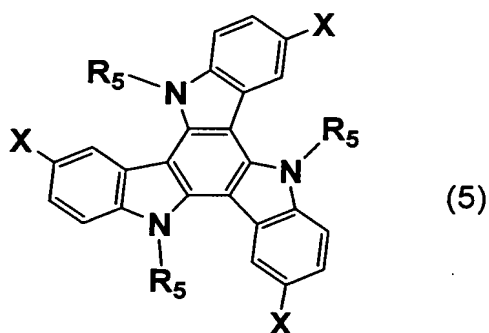
The compound obtained by the present invention has a fluorescent spectrum from 400 nm to around 700 nm and accordingly it is expected that the compound is usable in organic

EL materials, solar cells, etc.

[0157] to [0159]

The present invention [4] lies in a process for producing a Sym-triindole derivative represented by the general
5 formula (7), which comprises reacting an N-substituted-5-halo-oxyindole represented by the general formula (4) with a phosphorus oxyhalide to obtain an N-substituted-5-halo-triindole represented by the following general formula (5)

[formula 36]



10 (wherein R₅ is C2-C12 alkyl group, substituted C2-C12 alkyl group, C2-C12 haloalkyl group or aryl C1-C6 alkyl group; and X is halogen) (first step) and reacting it with a boric acid compound represented by the general formula (6) (second step).

[0160]

15 Firstly, the first step of the present invention is de-

scribed. The present step is a step for reacting an N-substituted-5-halo-oxyindole represented by the general formula (4) with a phosphorus oxyhalide to produce an N-substituted-5-halo-triindole.

5 [0161]

The molar ratio of the N-substituted-5-halo-oxyindole represented by the general formula (4) and the phosphorus oxyhalide both used as raw materials in the present step may be any molar ratio as long as stirring is possible. However,
10 the amount of the phosphorus oxyhalide is preferably 0.5 to 30 liters, more preferably 1 to 10 liters relative to 1 mole of the oxyindole.

[0162]

The temperature of the reaction is not particularly restricted as long as the reaction proceeds at the temperature,
15 but is preferred to be, for example, a range from 80°C to the reflux temperature.

[0163]

The period of the reaction is not particularly restricted but is preferred to be, for example, 1 to 48 hours.
20

[0164]

Then, the second step is described. The present step is a step for reacting the N-substituted-5-halo-oxyindole obtained in the previous step, with a boric acid compound represented by the general formula (6) to produce a Sym-
5 triindole derivative represented by the general formula (7).

[0165]

The boric acid compound represented by the general formula (6) (which is a raw material in the present step) is explained. In the general formula (6), R_6 is hydrogen atom,
10 formyl group, cyano group, or C1-C6 alkoxy carbonyl group such as methoxycarbonyl, ethoxycarbonyl, n-propoxycarbonyl, isopropoxycarbonyl or the like; and R_a and R_b are each independently hydrogen atom, C1-C6 alkyl group, or optionally substituted phenyl and may be combined to each other to form a ring.
15 (Therefore, R_a-B-R_b indicates formic acid group; straight chain or branched chain C1-C12 alkyl borate such as methoxyboric acid, ethoxyboric acid, isopropoxyboric acid or the like; or cyclic boric acid ester such as 4,4,5,5-tetramethyl-
20 1,3,2-dioxaborolane-2-yl, 5,5-dimethyl-1,3,2-dioxaboroxan-2-

yl or the like.

Therefore, as specific examples of the boric acid compound represented by the general formula (6), usable in the second step reaction, there can be mentioned phenylboric acid, methyl phenylborate, ethyl phenylborate, isopropyl phenylborate, 2-formylphenylboric acid, 3-formylphenylboric acid, 4-formylphenylboric acid, methyl 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolane-2-yl)benzoate, ethyl 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolane-2-yl)benzoate, 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolane-2-yl)benzonitrile, and 4-(5,5-dimethyl-1,3,2-dioxaboroxan-2-yl)benzaldehyde.

[0166]

In the reaction of the second step, the molar ratio of the boric acid compound represented by the general formula (6) and the N-substituted-5-halo-triindole derivative may be any molar ratio; however, the preferred amount of the former is, for example, 3 to 15 moles relative to 1 mole of the N-substituted-5-halo-triindole derivative.

[0167]

The reaction is conducted using a solvent. As to the

solvent usable therein, there is no restriction as long as the solvent does not impair the present reaction. There are preferably used, for example, aromatic hydrocarbons such as benzene, toluene, xylene and the like; aprotic polar solvents
5 such as dimethylformamide, diethylformamide, dimethylacetamide and the like; cyclic ethers such as tetrahydrofuran (THF), dioxane and the like; chain ethers such as monoglyme, diglyme and the like; alcohols such as methanol, ethanol, butanol, ethylene glycol, propylene glycol, PEG-400
10 and the like; and water.

[0168]

The solvent used in the reaction may be a single solvent or a mixed solvent of any mixing ratio.

[0169]

15 The amount of the solvent used in the reaction may be any as long as stirring is possible; however, it is preferred to be, for example, 0.1 to 10 liters relative to 1 mole of the Sym-N-substituted-5-halo-triindole (a raw material).

[0170]

20 The temperature of the reaction is not particularly re-

stricted as long as the reaction proceeds at the temperature,
but is preferred to be, for example, 40°C to the reflux tem-
perature of the solvent used. The period of the reaction is
not particularly restricted but is, for example, 0.5 to 48
5 hours.

[0171]

The invention [5] is the second step of the invention
[4] described above.

[0172]

10 The invention [6] is the first step of the invention
[4] described above.

[0173]

The invention [7] lies in a process for producing a
Sym-triindole derivative represented by the general formula
15 (10), which comprises reacting a Sym-triindole derivative
represented by the general formula (8) with a methylene com-
pound represented by the general formula (9).

[0174]

In the methylene compound represented by the general
20 formula (9), used in the reaction of the invention [7], R_8 is

hydrogen atom or cyano group; and R₉ is cyano group, carboxylic acid group (e.g. carboxylic acid, sodium carboxylate, potassium carboxylate, lithium carboxylate or pyridinium carboxylate), C1-C6 alkoxycarbonyl group (e.g. methoxycarbonyl or ethoxycarbonyl), aryloxycarbonyl group (e.g. phenoxycarbonyl group), aryl group (e.g. phenyl group, naphthyl group, 2-pyridyl group, 3-pyridyl group, 4-pyridyl group, 2-pyrimidine group, 4-pyrimidine group or triazine group), or substituted aryl group (e.g. 2-nitrophenyl group, 3-nitrophenyl group, 4-nitrophenyl group, 4-formylphenyl group, 4-cyanophenyl group or 6-methyl-2-pyridine group). The substituted aryl group of R₉ includes a salt thereof when the group is a heterocyclic ring containing nitrogen as heterocyclic atom, such as pyridine or the like.

[0175]

Therefore, as the methylene compound represented by the general formula (9), usable in the reaction of the invention [7], there can be mentioned, for example, acetonitrile, malononitrile, methyl acetate, ethyl acetate, cyanoacetic acid, methyl cyanoacetate, ethyl cyanoacetate, phenyl

cyanoacetate, 4-nitrophenyl cyanoacetate, 2-
cyanomethylpyridine, 3-cyanomethylpyridine, 4-
cyanomethylpyridine, 2-cyanomethylpyrimidine, benzyl cyanide
and 4-nitrobenzyl cyanide.

5 [0176]

The molar ratio of the methylene compound represented
by the general formula (9), usable in the reaction of the in-
vention [7] may be any molar ratio as long as the reaction is
not impaired. However, the amount of the methylene compound
10 is 0.1 to 30 moles, preferably 1 to 10 moles relative to 1
mole of the Sym-triindole derivative represented by the gen-
eral formula (8).

[0177]

The reaction is ordinarily conducted using a solvent.
15 As the solvent used in the reaction, there can be mentioned,
for example, aprotic polar solvents such as dimethylformamide,
dimethylacetamide, diethylacetamide, N-methylpyrrolidone,
N,N-dimethylimidazolinone, tetramethylurea and the like; aro-
matic hydrocarbons such as toluene, xylene, chlorobenzene and
20 the like; chain or cyclic ethers such as diethyl ether,

diisopropyl ether, tetrahydrofuran, dioxane and the like; acetic acid esters such as methyl acetate, ethyl acetate, butyl acetate and the like; and nitriles such as acetonitrile and the like. The solvent used in the reaction may be a single solvent or a mixed solvent of any mixing ratio.

[0178]

The amount of the solvent used in the reaction may be any as long as stirring is possible. However, the preferred amount is, for example, 0.1 to 30 liters, preferably 0.5 to 10 liters relative to 1 mole of the Sym-triindole derivative of the general formula (8) which is a raw material.

[0179]

The temperature of the reaction may be any temperature as long as the reaction proceeds at the temperature, but is preferred to be, for example, -20°C to the reflux temperature of the solvent used.

[0180]

The period of the reaction is not particularly restricted but is preferred to be, for example, 1 to 48 hours.

[0181]

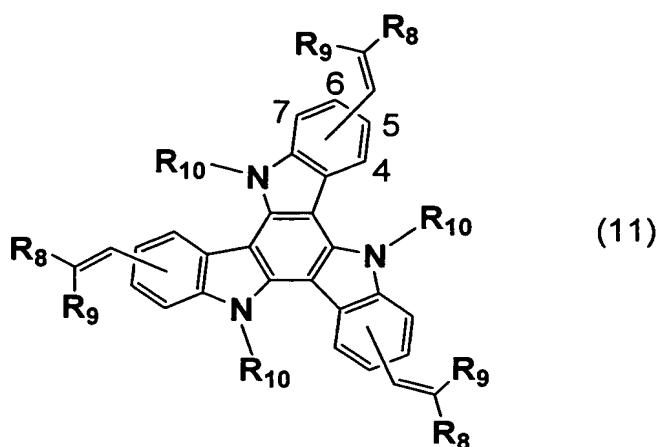
The invention [8] lies in a Sym-vinyltriindole derivative represented by the general formula (11). In the general formula (11), R₈ is hydrogen atom or cyano group; and R₉ is cyano group, carboxylic acid group (e.g. carboxylic acid, sodium carboxylate, potassium carboxylate, lithium carboxylate or pyridinium carboxylate), C1-C6 alkoxycarbonyl group (e.g. methoxycarbonyl or ethoxycarbonyl), arylcarbonyl group (e.g. phenoxycarbonyl), aryl group (e.g. phenyl group, naphthyl group, 2-pyridyl group, 3-pyridyl group, 4-pyridyl group, 2-pyrimidine group, 4-pyrimidine group or triazine group), or substituted aryl group (e.g. 2-nitrophenyl group, 3-nitrophenyl group, 4-nitrophenyl group, 4-formylphenyl group, 4-cyanophenyl group or 6-methyl-2-pyridine group).

[0182]

Therefore, compounds shown in Table 3 can be mentioned as the compound of the present invention.

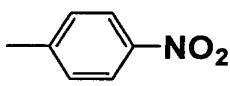
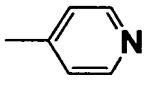
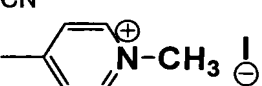
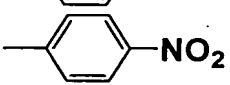
[0183]

[Formula 37]



[0184]

Table 3

Compound No.	Substitution position (each of three)	R ₈	R ₉	R ₁₀	Melting point (°C)
43	5	CN	CN	n-Hex	254—258
44	5	CN	COOEt	n-Hex	145—153
45	5	CN		n-Hex	300 or higher
46	5	CN		n-Hex	255—260
47	5	CN	CN	(2-Et)Hex	290(decomposed)
48	4	CN	CN	n-Hex	
49	6	CN	CN	n-Hex	
50	7	CN	CN	n-Hex	
51	5	CN		n-Hex	236—238
52	5	CN		(2-Et)Hex	267—272
53	5	CN	COOH	(2-Et)Hex	298—301

[0185]

The invention [9] lies in a process for producing a

5 Sym-triindole derivative represented by the general formula

(11), which comprises reacting an oxyindole compound represented by the general formula (12) with a phosphorus oxyhalide to obtain a Sym-halo-triindole derivative represented by the general formula (13) (first step), reacting it with a
5 formylating agent in the presence of butyllithium to obtain a Sym-formyltriindole derivative represented by the general formula (14) (second step), and reacting it with a methylene compound represented by the general formula (9) (third step).
[0186]

10 Firstly, description is made on the first step of the present invention, that is, the reaction of oxyindole with phosphorus oxyhalide.

[0187]

Firstly, explanation is made on the oxyindole compound
15 represented by the general formula (12) used as a raw material.

[0188]

In the general formula (12), R_{10} is, for example, a straight chain or branched chain C2-C12 alkyl group such as
20 ethyl group, n-propyl group, n-butyl group, n-pentyl group,

n-hexyl group, isopropyl group, 2-ethylhexyl group or the like; or a straight chain or branched chain C2-C12 haloalkyl group such as chloroethyl group, dichloroethyl group, trifluoroethyl group or the like.

5 [0189]

Therefore, as the oxyindole compound represented by the general formula (12), usable in the present invention, there can be mentioned, for example, N-ethyl-4-bromooxyindole, N-ethyl-5-bromooxyindole, N-ethyl-6-bromooxyindole, N-ethyl-7-bromooxyindole, N-(n-propyl)-5-bromooxyindole, N-(n-butyl)-5-bromooxyindole, N-(n-pentyl)-5-bromooxyindole, N-(n-hexyl)-5-bromooxyindole, N-(isopropyl)-5-bromooxyindole, and N-(2-ethylhexyl)-5-bromooxyindole.

[0190]

15 The oxyindole used in the present invention is known or can be produced by reacting a corresponding aniline with chloroacetyl chloride in the presence of aluminum chloride.

[0191]

The molar ratio of the oxyindole used as a raw material and the phosphorus oxyhalide may be any molar ratio as long

as stirring is possible. However, the preferred amount of the phosphorus oxyhalide is, for example, 0.5 to 30 liters, preferably 1 to 10 liters relative to 1 mole of the oxyindole.

[0192]

5 As the phosphorus oxyhalide, there can be mentioned, for example, phosphorus oxychloride and phosphorus oxybromide. Ordinarily, phosphorus oxychloride is used preferably.

[0193]

 The temperature of the reaction may be any temperature
10 as long as the reaction proceeds at the temperature, but is preferred to be, for example, 80°C to the reflux temperature.

[0194]

 The period of the reaction is not particularly restricted but is preferred to be, for example, 1 to 48 hours.

15 [0195]

 Next, description is made on the formylation of the second step.

[0196]

 The second step of the present invention is a step for
20 reacting the Sym-halo-indole represented by the general for-

mula (13) with a formylating agent in the presence of butyllithium to produce a Sym-formyltriindole represented by the general formula (14).

[0197]

5 As the formylating agent used in the present invention, there can be mentioned, for example, dimethylformamide, diethylformamide, formamide, N-formylpiperidine and N-formylpyrrolidine. Dimethylformamide and N-formylpiperidine are preferred. The molar ratio of the formylating agent and
10 the Sym-halo-triindole derivative of the general formula (13) (a raw material) may be any molar ratio as long as the reaction is not impaired; however, the amount of the formylating agent is, for example, 3 to 30 moles, preferably 3 to 9 moles relative to 1 mole of the Sym-halo-triindole derivative represented by the general formula (13).
15

[0198]

Butyllithium is simultaneously used in the present invention. The molar ratio of the butyllithium used in the present invention may be any molar ratio as long as the reaction is not impaired. However, the amount of the butyllith-
20

ium is, for example, 3 to 15 moles, preferably 3 to 6 moles relative to 1 mole of the Sym-halo-triindole derivative represented by the general formula (13).

[0199]

5 The present step is generally conducted using a solvent. As the solvent used in the present step, there can be mentioned, for example, aromatic hydrocarbons such as toluene, xylene and the like; chain or cyclic ethers such as diethyl ether, diisopropyl ether, tetrahydrofuran (THF), dioxane and
10 the like; and hexamethylphosphoric triamide (HMPA). Tetrahydrofuran (THF) is preferred. The solvent used in the reaction may be a single solvent or a mixed solvent of any mixing ratio.

[0200]

15 The amount of the solvent used in the reaction may be any amount as long as stirring is possible, but is preferred to be, for example, 0.1 to 10 liters relative to 1 mole of the Sym-halo-triindole derivative which is a raw material.

[0201]

20 The temperature of the reaction may be any temperature

as long as the reaction proceeds at the temperature, but is preferred to be, for example, -100°C to the reflux temperature of the solvent used, preferably -78 to 30°C .

[0202]

5 The period of the reaction is not particularly restricted but is, for example, 0.5 to 48 hours.

[0203]

 The third step is a step for reacting the Symformyltriindole derivative represented by the general formula
10 (14), obtained in the second step, with a methylene compound represented by the general formula (9) to produce a Symtriindole derivative represented by the general formula (11).

[0204]

 In the methylene compound represented by the general
15 formula (9), used in the reaction of the third step, R_8 is hydrogen atom or cyano group; and R_9 is cyano group, carboxylic acid group (e.g. carboxylic acid, sodium carboxylate, potassium carboxylate, lithium carboxylate or pyridinium carboxylate), C1-C6 alkoxy carbonyl group (e.g. methoxycarbonyl
20 group or ethoxycarbonyl group), aryloxy carbonyl group (e.g.

phenoxycarbonyl group), aryl group [e.g. phenyl group, naphthyl group, or aromatic heterocyclic ring group of single ring or condensed ring containing nitrogen, oxygen or sulfur as heterocyclic atom, such as 2-pyridyl group, 3-pyridyl group, 4-pyridyl group, 2-pyrimidine group, 4-pyrimidine group, triazine group, pyrazolyl group, triazolyl group, quinolinyl group, indolyl group, isoindolyl group, thienyl group, benzothienyl group, furyl group, benzofuryl group, thiazolyl group, benzothiazolyl group, oxazolyl group, isoxazolyl group, benzoxazolyl group, thiadiazolyl group, benzothiadiazolyl group or the like], or substituted aryl group (e.g. 2-nitrophenyl group, 3-nitrophenyl group, 4-nitrophenyl group, 4-formylphenyl group, 4-cyanophenyl group or 6-methyl-2-pyridine group).

[0205]

Therefore, as the methylene compound represented by the general formula (9), usable in the reaction of the third step, there can be mentioned, for example, acetonitrile, malononitrile, methyl acetate, ethyl acetate, cyanoacetic acid, methyl cyanoacetate, ethyl cyanoacetate, phenyl cyanoacetate,

(4-nitrophenyl) cyanoacetate, 2-cyanomethylpyridine, 3-cyanomethylpyridine, 4-cyanomethylpyridine, 2-cyanomethylpyridine, benzyl cyanide and 4-nitrobenzyl cyanide.

The molar ratio of the methylene compound represented by the
5 general formula (9), used in the reaction may be any molar
ratio as long as the reaction is not impaired. However, the
amount of the methylene compound is 0.1 to 30 moles, preferably
1 to 10 moles relative to 1 mole of the Sym-
formyltriindole derivative represented by the general formula
10 (14).

[0206]

The reaction is ordinarily conducted using a solvent.
As the solvent used in the reaction, there can be mentioned,
for example, aprotic polar solvents such as dimethylformamide,
15 dimethylacetamide, diethylacetamide, N-methylpyrrolidone,
N,N-dimethylimidazolinone, tetramethylurea and the like; aromatic
hydrocarbons such as toluene, xylene, chlorobenzene and
the like; chain or cyclic ethers such as diethyl ether,
diisopropyl ether, tetrahydrofuran (THF), dioxane and the
20 like; acetic acid esters such as methyl acetate, ethyl ace-

tate, butyl acetate and the like; and nitriles such as acetonitrile and the like. The solvent used in the reaction may be a single solvent or a mixed solvent of any mixing ratio.

[0207]

5 The amount of the solvent used in the reaction may be any as long as stirring is possible. However, the preferred amount is, for example, 0.1 to 30 liters, preferably 0.5 to 10 liters relative to 1 mole of the Sym-formyltriindole derivative of the general formula (14) which is a raw material.

10 [0208]

 The temperature of the reaction may be any temperature as long as the reaction proceeds at the temperature, but is preferred to be, for example, -20°C to the reflux temperature of the solvent used.

15 [0209]

 The period of the reaction is not particularly restricted but is preferred to be, for example, 1 to 48 hours.

[0210]

 The invention [10] is described. This invention is the
20 third step of the invention [9].

[0211]

The invention [11] is described. This invention is the second step of the invention [9].

[0212]

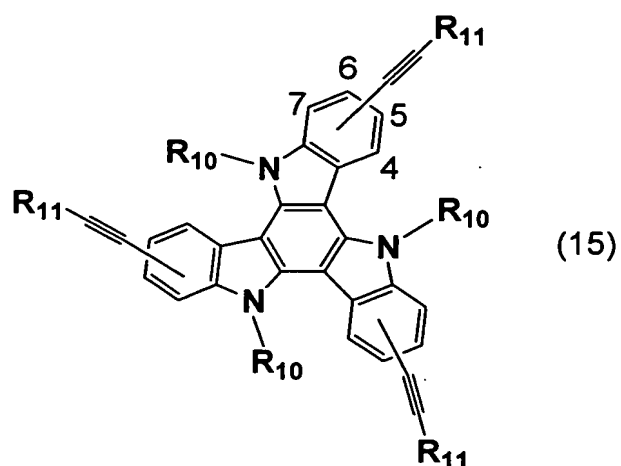
5 The invention [12] is described. This invention lies in a Sym-triindole derivative represented by the general formula (15). In the general formula (15), R_{10} has the same definition as given above; and R_{11} is, for example, aryl group (e.g. phenyl group, 2-pyridyl group, 3-pyridyl group, 10 4-pyridyl group or 2-pyrimidine group), or substituted aryl group [e.g. 4-nitrophenyl group, 4-cyanophenyl group, or 4-(methoxycarbonyl)phenyl group].

[0213]

Therefore, for example, compounds shown in Table 4 can
15 be mentioned as the compound of the present invention.

[0214]

[Formula 38]



[0215]

Table 4

Compound No.	Substitution position (each of three)	R ₁₀	R ₁₁	Melting point (°C)
54	5	n-Hex	Ph	180—189
55	4	n-Hex	Ph	
56	6	n-Hex	Ph	
57	7	n-Hex	Ph	
58	5	(2-Et)Hex	Ph	
59	5	n-Hex	4-Py	
60	5	(2-Et)Hex	4-Py	

[0216]

The invention [13] is described.

5 [0217]

The present invention lies in a process for producing a Sym-triindole derivative represented by the general formula (15), which comprises reacting a Sym-halo-triindole derivative represented by the general formula (13), with an acety-

lene derivative represented by the general formula (16).

[0218]

In the general formula (16), R_{11} is, for example, aryl group (e.g. phenyl group, 2-pyridyl group, 3-pyridyl group, 5 4-pyridyl group or 2-pyrimidine group), or substituted aryl group [e.g. 4-nitrophenyl group, 4-cyanophenyl group or 4-(methoxycarbonyl)phenyl group]; and R_{12} is hydrogen atom or trimethylsilyl group.

[0219]

10 Therefore, as the acetylene derivative of the general formula (16) used in the reaction, there can be mentioned, for example, phenylacetylene, 1-phenyl-2-trimethylsilylacetylene, (pyridin-2-yl)acetylene, (pyridin-3-yl)acetylene, (pyridin-4-yl)acetylene, (4-15 cyanophenyl)acetylene, and 1-(pyridin-4-yl)-2-trimethylsilylacetylene.

[0220]

In the reaction, the molar ratio of the Sym-halo-triindole derivative represented by the general formula (13) 20 and the acetylene derivative represented by the general for-

mula (16) may be any molar ratio. However, the amount of the acetylene derivative represented by the general formula (16) is preferably 3 to 10 moles per 1 mole of the Sym-halo-triindole derivative represented by the general formula (13).

5 [0221]

The reaction is conducted generally using a solvent. As the solvent usable in the reaction, there can be mentioned, for example, aromatic hydrocarbons such as toluene, xylene, chlorobenzene and the like; halogenated aliphatic hydrocarbons such as dichloromethane, chloroform and the like; 10 aprotic polar solvents such as dimethylformamide, diethylformamide, N-methylpyrrolidone, dimethylacetamide and the like; chain or cyclic ethers such as diethyl ether, tetrahydrofuran (THF), dioxane and the like; alcohols such as methanol, ethanol 15 and the like; polyethylene glycols such as PEG-400 and the like; nitriles such as acetonitrile and the like; nitrogen-containing bases such as pyrrolidine, triethylamine, pyridine and the like; and water. The solvent used in the present reaction may be a single solvent or a mixed solvent 20 of any mixing ratio.

[0222]

The amount of the solvent used in the reaction may be any amount as long as stirring is possible; however, the amount is preferably 0.1 to 20 liters relative to 1 mole of
5 the Sym-halo-triindole derivative of the general formula (13) which is a raw material.

[0223]

The temperature of the reaction may be any temperature as long as the reaction proceeds at the temperature, but is
10 preferred to be, for example, 20°C to the reflux temperature of the solvent used.

[0224]

The period of the reaction is not particularly restricted but is preferred to be, for example, 1 to 48 hours.

15 [0225]

The invention [14] is described. This invention is the first step of the invention [9]

Examples

20 [0226]

Next, the processes for producing the present invention compounds are specifically described by way of Examples. However, the present invention is in no way restricted by these Examples.

5 [0227]

Example 1: Production of Sym-N-(n-hexyl)-5-bromotriindole

In a 500-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed 40 g of N-(n-hexyl)-5-bromooxyindole and 240 ml of
10 phosphorus oxychloride. Refluxing was conducted with heating, for 8 hours, to give rise to a reaction. After the completion of the reaction, phosphorus oxychloride was recovered under reduced pressure. To the resulting residue were added water and a sodium hydroxide aqueous solution under cooling.
15 The resulting crystals were collected by filtration and dissolved in 300 ml of toluene. The resulting solution was hot-filtered to remove the insolubles. The filtrate (toluene layer) was washed with water and then concentrated. The concentrate (residue) was recrystallized (toluene : ethyl acetate = 1 : 9) to obtain 10.9 g of Sym-N-(n-hexyl)-5-
20

bromotriindole. Yield: 29%

[0228]

Melting point: 205-207°C

FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 835

5 ^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.84 (t, $J=6.6$ Hz, 9H), 1.20-1.41 (m, 18H),

1.88-2.20 (br, 6H), 4.59 (t, $J=8.1$ Hz, 6H),

7.41 (t, $J=8.7$ Hz, 3H), 7.52 (dd, $J=1.5, 8.7$ Hz, 3H),

8.21 (d, $J=1.5$ Hz, 3H)

10 [0229]

Example 2: Production of Sym-N-(n-hexyl)-5-phenyltriindole

In a 50-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed, in an argon atmosphere, 165 mg of Sym-N-(n-hexyl)-5-bromotriindole, 35 mg of tetrakis(triphenylphosphine) palladium (O) and 20 ml of toluene. Thereto were added a solution of 109 mg of phenylboronic acid in 5 ml of ethanol and 10 ml of a saturated sodium bicarbonate aqueous solution. Refluxing was conducted with heating, for 5 hours, to give rise to a reaction. Aster the completion of the reaction, the system

was cooled to room temperature and filtered. To the resulting filtrate were added 50 ml of ethyl acetate and 30 ml of a saturated sodium chloride aqueous solution, for layer separation. The resulting ethyl acetate layer was concentrated.

5 The concentrate was subjected to silica gel column chromatography (chloroform) to obtain 39 mg of Sym-N-(n-hexyl)-5-phenyltriindole. Yield: 23.8%

[0230]

FAB-Mass (NBA, positive): $[(M+H)^+]$ = 826

10 $^1\text{H-NMR}$ (270 MHz, CDCl_3), δ (ppm):

0.74 (t, $J=7.1$ Hz, 9H), 1.10-1.25 (m, 18H),

2.00-2.10 (br, 6H), 4.96 (t, $J=7.9$ Hz, 6H),

7.30-7.41 (m, 3H), 7.48-7.55 (m, 6H),

7.64-7.78 (m, 12H), 8.49 (s, 3H)

15 [0231]

Example 3: Production of Sym-N-(n-hexyl)-5-(4-formylphenyl)triindole

In Example 2, phenylboronic acid was replaced by 150 mg of 4-formylphenylboronic acid, whereby was obtained 35 mg of
20 Sym-N-(n-hexyl)-5-(4-formylphenyl)triindole. Yield: 19.3%

[0232]

Melting point: 245-257°C

FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 910

$^1\text{H-NMR}$ (270 MHz, CDCl_3), δ (ppm):

5 0.73 (t, $J=7.1$ Hz, 9H), 1.00-1.30 (m, 18H),
 2.00-2.20 (br, 6H), 4.98 (t, $J=8.1$ Hz, 6H),
 7.74 (s, 6H), 7.91 (d, $J=8.2$ Hz, 6H),
 8.03 (d, $J=8.6$ Hz), 8.54 (s, 3H), 10.10 (s, 3H)

[0233]

10 Example 4: Production of Sym-N-(n-hexyl)-5-(4-ethoxycarbonylphenyl)triindole

 In Example 2, phenylboronic acid was replaced by 247 mg
 of ethyl 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolane-2-yl)benzoate, whereby was obtained by 78 mg of Sym-N-(n-
15 hexyl)-5-(4-ethoxycarbonylphenyl)triindole. Yield: 37.7%

[0234]

Melting point: 203-213°C

FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 1042

$^1\text{H-NMR}$ (270 MHz, CDCl_3), δ (ppm):

20 0.74 (t, $J=7.1$ Hz, 6H), 1.02-1.28 (m, 18H),

1.45 (t, J=6.9, 7.2 Hz, 9H), 2.00-2.18 (br, 6H),
4.44 (q, J=6.9, 7.2 Hz, 6H), 4.92 (t, J=7.9 Hz, 6H),
7.68-7.74 (m, 6H), 7.80 (d, J=8.2 Hz, 6H),
8.19 (d, J=8.2 Hz, 6H), 8.50 (s, 3H)

5 [0235]

Example 5: Production of Sym-N-(n-hexyl)-5-(4-cyanophenyl)triindole

In Example 2, phenylboronic acid was replaced by 206 mg
of 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolane-2-
10 yl)benzonitrile, whereby was obtained 21.5 mg of Sym-N-(n-hexyl)-5-(4-cyanophenyl)triindole. Yield: 11.9%

[0236]

Melting point: 297-303°C

FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 901

15 $^1\text{H-NMR}$ (270 MHz, CDCl_3), δ (ppm):

0.74 (t, J=7.1 Hz, 9H), 1.05-1.25 (m, 18H),
1.98-2.11 (br, 6H), 4.94 (t, J=7.5 Hz, 6H),
7.69 (d, J=8.1 Hz, 3H), 7.73 (d, J=8.4 Hz, 3H),
7.79 (d, J=8.7 Hz, 6H), 7.83 (d, J=8.6 Hz, 6H),
20 8.46 (s, 3H)

[0237]

Example 6: Production of Sym-N-(n-hexyl)-5-(4-(2,2-dicyanovinyl)phenyl)triindole

In a 50-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed, in an argon atmosphere, 50 mg of Sym-N-(n-hexyl)-5-(4-formylphenyl)triindole, 16 mg of malononitrile and 5.5 ml of THF. Thereto were added 0.165 mg of acetic acid and 2.3 mg of pyrrolidine. Stirring was conducted at room temperature for 4 hours to give rise to a reaction. After the completion of the reaction, the reaction mixture was concentrated. The concentrate (residue) was subjected to silica gel column chromatography (chloroform) to obtain 25 mg of Sym-N-(n-hexyl)-5-(4-(2,2-dicyanovinyl)phenyl)triindole. Yield: 43.2%

[0238]

Melting point: 281-291°C

FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 1054

[0239]

Example 7: Production of Sym-N-(n-hexyl)-5-methyltriindole

In a 100-ml, three-necked flask provided with a mag-

netic stirrer, a reflux condenser and a thermometer were placed 7.60 g of N-(n-hexyl)-5-methyloxyindole and 60 ml of phosphorus oxychloride. They were subjected to a reaction at 100°C for 11 hours. After the completion of the reaction, 5 phosphorus oxychloride was recovered. Water was added and neutralization was made with sodium hydroxide. The resulting crystals were collected by filtration and dissolved in 250 ml of toluene. The toluene solution was filtered to remove the insolubles. The filtrate was washed with water, dried with 10 sodium sulfate, and concentrated, after which recrystallization (toluene : ethyl acetate = 1 : 9) was conducted to obtain 2.16 g of Sym-N-(n-hexyl)-5-methyltrindole. Yield: 31% [0240]

Melting point: 160°C

15 EI-Mass: (M^+) = 639

$^1\text{H-NMR}$ (300 MHz, CDCl_3), δ (ppm):

0.83 (t, $J=6.6$ Hz, 9H), 1.20-1.45 (m, 18H),
2.03 (quint, $J=7.8$ Hz, 6H), 2.61 (s, 9H),
4.83 (t, $J=8.1$ Hz, 6H), 7.25 (bd, $J=8.4$ Hz, 3H),
20 7.50 (d, $J=8.4$ Hz, 3H), 8.06 (bs, 3H)

[0241]

Example 8: Production of Sym-N-(n-hexyl)-5-formyltriindole

In a 300-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed, in a nitrogen atmosphere, 1.5 g of Sym-N-(n-hexyl)-5-bromotriindole and 180 ml of tetrahydrofuran (THF). They were cooled to -78°C . Thereto was dropwise added 6.9 ml of 1.57 M n-butyllithium slowly and a reaction was allowed to take place for 1 hour at the same temperature. Then, 1.2 g of N-formylpiperidine was added slowly and a reaction was allowed to take place at -78°C for 1 hour. The reaction mixture was heated to 0°C and diluted hydrochloric acid was added for hydrolysis. Then, extraction with toluene was conducted. The toluene layer was dried with sodium sulfate and concentrated. The concentrate (residue) was recrystallized (tetrahydrofuran, THF) to obtain 1.01 g of Sym-N-(n-hexyl)-5-formyltriindole. Yield: 82%

[0242]

Melting point: $185-187^{\circ}\text{C}$

FAB-Mass (NBA, Positive): $(\text{M}^+) = 681$

IR (KBr, cm^{-1}): 1687

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.80 (t, $J=7.2$ Hz, 9H), 1.20-1.40 (m, 18H),
2.03 (br, 6H), 4.93 (t, $J=7.8$ Hz, 6H),
5 7.72 (d, $J=8.7$ Hz, 3H), 8.02 (dd, $J=0.9, 8.4$ Hz, 3H),
8.77 (d, $J=1.2$ Hz, 3H), 10.17 (s, 3H)

Elemental analysis:

Calculated for $\text{C}_{45}\text{H}_{51}\text{N}_3\text{O}_3$ (%): C 79.26, H 7.54, N 6.16

Measured: C 79.20, H 7.49, N 6.19

10 [0243]

Example 9: Production of Sym-N-(n-hexyl)-5-(2,2-dicyanovinyl)triindole

In a 50-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed
15 1.00 g of Sym-N-(n-hexyl)-5-formyltriindole, 0.437 g of malononitrile, 62.4 mg of piperidine, 4.40 mg of acetic acid and 10 ml of tetrahydrofuran (THF). A reaction was conducted at 50°C for 3 hours. After the completion of the reaction, the solvent was removed by distillation. The resulting resi-
20 due was recrystallized using tetrahydrofuran (THF) to obtain

0.681 g of Sym-N-(n-hexyl)-5-(2,2-dicyanovinyl)triindole.

Yield: 56%

[0244]

Melting point: 254-258°C

5 FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 826

IR (KBr, cm^{-1}): 2223

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.69 (t, $J=7.2$ Hz, 9H), 1.05-1.30 (m, 18H),

1.82 (br, 6H), 5.09 (t, $J=7.2$ Hz, 6H),

10 7.79 (d, $J=8.7$ Hz, 3H), 7.94 (s, 3H),

7.97 (dd, $J=1.0, 8.7$ Hz, 3H), 8.95 (d, $J=1.0$ Hz, 3H)

Elemental analysis:

Calculated for $\text{C}_{54}\text{H}_{51}\text{N}_9$ (%): C 78.52, H 6.22, N 15.26

Measured: C 78.30, H 6.26, N 15.16

15 [0245]

Example 10: Production of Sym-N-(n-hexyl)-(2-cyano-2-ethoxycarbonylvinyl)triindole

In a 50-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed 100
20 mg of Sym-N-(n-hexyl)-5-formyltriindole, 74.8 mg of ethyl

cyanoacetate, 6.3 mg of piperidine and 15 m of tetrahydrofuran (THF). A reaction was conducted at 50°C for 12 hours. After the completion of the reaction, the solvent was removed by distillation. The residue was recrystallized using tetrahydrofuran (THF) to obtain 99.5 mg of Sym-N-(n-hexyl)-(2-cyano-2-ethoxycarbonylvinyl)triindole. Yield: 70%

[0246]

Melting point: 145-153°C

FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 967

10 IR (KBr, cm^{-1}): 1590

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.68 (t, $J=7.2$ Hz, 9H), 1.05-1.18 (m, 18H),
1.45 (t, $J=7.2$ Hz, 9H), 1.78-1.91 (bs, 6H),
4.44 (q, $J=7.2$ Hz, 6H), 5.10 (t, $J=8.1$ Hz, 6H),
15 7.76 (d, $J=8.7$ Hz, 3H), 8.13 (dd, $J=1.0, 8.7$ Hz, 3H),
8.49 (s, 3H), 8.99 (d, $J=1.0$ Hz, 3H)

[0247]

Example 11: Production of Sym-N-(n-hexyl)-[2-cyano-2-(4-nitrophenyl)vinyl]triindole

20 In a 50-ml, three-necked flask provided with a magnetic

stirrer, a reflux condenser and a thermometer were placed 100 mg of Sym-N-(n-hexyl)-5-formyltriindole, 107 mg of 4-nitrobenzyl cyanide, 6.3 mg of piperidine and 15 ml of tetrahydrofuran (THF). A reaction was conducted at 50°C for 12 hours. After the completion of the reaction, the resulting crystals were collected by filtration to obtain 117 mg of Sym-N-(n-hexyl)-[2-cyano-2-(4-nitrophenyl)vinyl]triindole.

[0248]

Melting point: 300°C or higher

10 FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 1114

IR (KBr, cm^{-1}): 2212, 1340

[0249]

Example 12: Production of Sym-N-(n-hexyl)-[2-cyano-2-(4-pyridyl)vinyl]triindole

15 In a 50-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed 100 mg of Sym-N-(n-hexyl)-5-formyltriindole, 120 mg of 4-cyanomethylpyridine, 6.3 mg of piperidine and 15 ml of tetrahydrofuran (THF). A reaction was conducted at 50°C for 4 hours. After the completion of the reaction, the reaction

mixture was subjected to silica gel column chromatography to obtain 131 mg of Sym-N-(n-hexyl)-[2-cyano-2-(4-pyridyl)vinyl]triindole. Yield: 91%

[0250]

5 Melting point: 255-260°C

FAB-Mass (NBA, Positive): $[(M+H)^+]$ = 983

IR (KBr, cm^{-1}): 2210

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.64 (t, $J=7.2$ Hz, 9H), 0.90-1.14 (m, 18H),
10 1.82 (br, 6H), 4.81 (t, $J=7.2$ Hz, 6H),
7.52 (d, $J=8.7$ Hz, 3H), 7.59-7.61 (m, 6H), 7.82 (s, 3H),
7.96 (dd, $J=1.0, 8.7$ Hz, 3H), 8.63 (bs, 3H),
8.71-8.73 (m, 6H)

[0251]

15 Example 13: Production of Sym-N-(2-ethylhexyl)-5-bromotriindole

In a 100-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed 5.0 g of N-(2-ethylhexyl)-5-bromooxyindole and 30 ml
20 of phosphorus oxychloride. A reaction was conducted at 100°C

for 11 hours. After the completion of the reaction, phosphorus oxychloride was recovered under reduced pressure. To the residue were added water and sodium hydroxide for neutralization. Then, extraction with toluene was conducted. The
5 toluene layer was concentrated and the concentrate was subjected to silica gel column chromatography to obtain 1.11 g of Sym-N-(2-ethylhexyl)-5-bromotriindole. Yield: 24%

EI-MS (M^+): 918

[0252]

10 Melting point: 161°C

$^1\text{H-NMR}$ (300 MHz, CDCl_3), δ (ppm):

0.26-0.33 (m, 9H), 0.56-0.65 (m, 9H),

0.74-1.04 (m, 24H), 1.61-1.72 (br, 3H),

4.07 (d, $J=7.2$ Hz, 6H), 7.42 (d, $J=8.7$ Hz, 3H),

15 7.58 (dd, $J=1.5, 8.7$ Hz, 3H), 7.75 (bs, 3H)

[0253]

Example 14: Production of Sym-N-(2-ethylhexyl)-5-formyltriindole

In a 100-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were
20

placed, in a nitrogen stream, 1.1 g of Sym-N-(2-ethylhexyl)-5-bromotriindole and 50 ml of tetrahydrofuran (THF). They were cooled to -78°C . Thereto were dropwise added 4.2 ml of 1.57 M n-butyllithium slowly, followed by aging at the same
5 temperature for 1 hour. Then, 0.74 g of N-formylpiperidine was added dropwise slowly. A reaction was conducted at -78°C for 1 hour. Thereafter, the system was heated to 0°C ; diluted hydrochloric acid was added for hydrolysis; and extraction with toluene was conducted. The toluene layer was concentrated and the concentrate was subjected to silica gel
10 column chromatography to obtain 0.80 g of Sym-N-(2-ethylhexyl)-5-formyltriindole. Yield: 87%

[0254]

Melting point: 243°C

15 FAB-Mass (NBA, Positive): (M^{+}) = 765

IR (KBr, cm^{-1}): 1684

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.33-0.39 (m, 9H), 0.52-0.60 (m, 9H),

0.61-0.96 (m, 24H), 1.74-1.88 (br, 3H),

20 4.67 (d, $J=6.3$ Hz, 6H), 7.62 (d, $J=8.4$ Hz, 3H),

8.06 (dd, J=1.2, 8.4 Hz, 3H), 8.51 (bs, 3H),

10.26 (s, 3H)

[0255]

Example 15: Production of Sym-N-(2-ethylhexyl)-5-(2,2-dicyanovinyl)triindole

In a 500-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed 0.80 g of Sym-N-(2-ethylhexyl)-5-formyltriindole, 0.31 g of malononitrile, 62.4 mg piperidine and 150 ml of tetrahydrofuran (THF). A reaction was conducted at 50°C for 3 hours. After the completion of the reaction, the reaction mixture was concentrated. The residue was recrystallized using tetrahydrofuran (THF) to obtain 0.15 g of Sym-N-(2-ethylhexyl)-5-(2,2-dicyanovinyl)triindole. Yield: 16%.

[0256]

Melting point: 290°C (decomposed)

FAB-Mass (NBA, Positive): $[(M+H)^+] = 910$

IR (KBr, cm^{-1}): 2224

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.43-0.56 (m, 9H), 0.62-0.70 (m, 9H),

0.72-1.11 (m, 24H), 1.86-2.20 (br, 3H),
4.99-5.60 (m, 6H), 7.76 (d, J=8.4Hz, 3H),
7.93 (s, 3H), 9.11 (bs, 3H)

[0257]

5 Example 16: Production of Sym-N-(n-hexyl)-5-(phenylethynyl)triindole

In a 100-ml, three-necked flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed, in a nitrogen stream, 100 mg of Sym-N-(n-hexyl)-5-bromotriindole, 8.35 mg of bistriphenylphosphine palladium
10 dichloride, 2.27 mg of copper (I) iodide, 4 ml of triethylamine and 4 ml of toluene. A reaction was conducted at 70°C for 9 hours. After the completion of the reaction, the reaction mixture was cooled to room temperature and neutralized
15 with diluted hydrochloric acid. Extraction with toluene was conducted. The organic layer was washed with water and a saturated sodium chloride aqueous solution. The toluene layer was concentrated and the concentrate was subjected to silica gel column chromatography to obtain 10 mg of Sym-N-(n-hexyl)-5-(phenylethynyl)triindole. Yield: 9.3%
20

[0258]

Melting point: 180-189°C

IR (KBr, cm^{-1}): 2927, 1565, 1494

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

5 0.758 (t, $J=7.2$ Hz, 9H), 1.15-1.50 (m, 18H),
 2.00-2.17 (m, 6H), 4.79-4.87 (m, 6H), 7.34-7.42 (m, 9H),
 7.56-7.66 (m, 12H), 8.418 (s, 3H)

[0259]

Example 17: Production of methyl iodide salt of Sym-N-(n-
10 hexyl)-5-[2-cyano-2-(4-pyridyl)vinyl]triindole

 In a 25-ml reaction flask provided with a magnetic
 stirrer, a reflux condenser and a thermometer were placed 100
 mg (0.10 mmol) of Sym-N-(n-hexyl)-5-[2-cyano-2-(4-
 pyridyl)vinyl]triindole and 2 ml of methyl iodide. Stirring
15 was conducted at room temperature for 12 hours. The reaction
 mixture was concentrated under reduced pressure. The residue
 was washed with hexane, toluene, ethyl acetate and tetrahy-
 drofuran. The resulting crude crystals were dissolved in a
 small amount of methanol. Thereto was added dichloromethane
20 to give rise to recrystallization, to obtain 114 mg of dark

brown crystals. Yield: 81%

[0260]

Melting point: 236-238°C

IR (KBr, cm^{-1}): 2214

5 ^1H -NMR (300 MHz, DMSO-d_6), δ (ppm):

0.44 (t, $J=6.9$ Hz, 9H), 0.72-0.98 (m, 24H),

1.71 (m, 6H), 4.34 (s, 9H), 5.22 (m, 6H),

8.27 (d, $J=9.3$ Hz), 8.39 (d, $J=9.0$ Hz, 3H),

8.44 (d, $J=6.6$ Hz), 9.01 (s, 3H),

10 9.04 (d, $J=6.9$ Hz, 6H), 9.11 (s, 3H)

[0261]

Example 18: Production of Sym-N-(2-ethylhexyl)-5-[2-cyano-2-(4-nitrophenyl)vinyl]triindole

In a 25-ml reaction flask provided with a magnetic
15 stirrer, a reflux condenser and a thermometer were placed 200
mg (0.26 mmol) of Sym-N-(2-ethylhexyl)-5-formyltriindole, 254
mg (1.56 mmol) of 4-cyanomethylnitrobenzene, 0.5 ml of
piperidine and 5 ml of tetrahydrofuran. Stirring was con-
ducted at 65°C for 12 hours. After confirmation of the dis-
20 appearance of the raw materials by TLC, water was added. The

resulting reddish brown crystals were collected by filtration and washed with toluene. The resulting cake was purified by silica gel column chromatography to obtain 20 mg of orange crystals. Yield: 6%

5 [0262]

Melting point: 267-272°C

IR (KBr, cm^{-1}): 854, 1340, 2210

^1H -NMR (300 MHz, CDCl_3), δ (ppm):

0.45-0.54 (m, 9H), 0.60-0.66 (m, 9H),
10 0.68-1.12 (m, 24H), 1.99 (m, 3H), 5.11 (m, 6H),
7.74 (d, $J=8.4$ Hz), 7.87-7.96 (m, 12H),
8.35 (d, $J=8.7$ Hz, 6H), 9.17 (s, 3H)

Elemental analysis:

Calculated for $\text{C}_{75}\text{H}_{75}\text{N}_9\text{O}_6$ (%): C 75.16, H 6.31, N 10.52

15 Measured: C 75.13, H 6.31, N 10.38

[0263]

Example 19: Production of Sym-N-(2-ethylhexyl)-5-[2-cyano-2-hydroxycarbonylvinyl]triindole

In a 50-ml reaction flask provided with a magnetic
20 stirrer, a reflux condenser and a thermometer were placed 371

mg (0.48 mmol) of Sym-N-(2-ethylhexyl)-5-formyltriindole, 618
mg (7.2 mmol) of cyanoacetic acid, 112 mg (1.44 mmol) of ammonium acetate and 20 ml of tetrahydrofuran (THF). Stirring was conducted at 65°C for 12 hours. Toluene was added. The
5 resulting yellow cake was collected by filtration and washed with water and toluene. The crude cake was dissolved in tetrahydrofuran. Toluene was added to give rise to recrystallization, and a cake was collected by filtration. The cake was subjected to silica gel column chromatography for purification.
10 Recrystallization from tetrahydrofuran was conducted to obtain 376 mg of yellow crystals.

[0264]

FAB-Mass (NBA, Positive): (M^+) = 967

Melting point: 298-301°C

15 IR (KBr, cm^{-1}): 1716, 2224, 2960

^1H -NMR (300 MHz, DMSO- d_6), δ (ppm):

0.32-0.39 (m, 9H), 0.51 (m, 9H), 0.60-0.97 (m, 24H),

1.77 (m, 3H), 4.99 (m, 6H), 8.07 (d, $J=8.7$ Hz, 3H),

8.62 (s, 3H), 8.96 (s, 3H), 13.7 (bs, 3H)

20 Elemental analysis:

Calculated for $C_{60}H_{66}N_6O_6$ (%): C 74.51, H 6.88, N 8.69

Measured: C 74.90, H 6.93, N 8.27

[0265]

Example 20: Production of Sym-N-(2-ethylhexyl)-5-
5 cyanotriindole

In a 200-ml reaction flask provided with a magnetic stirrer, a reflux condenser and a thermometer were placed 150 mg (0.20 mmol) of Sym-N-(2-ethylhexyl)-5-formyltriindole, 50 mg of hydroxylamine hydrochloride, 60 mg of sodium acetate,
10 50 ml of methanol and 50 ml of toluene. Stirring was conducted at room temperature for 12 hours to give rise to a reaction. After the completion of the reaction, the solvent was removed by distillation under reduced pressure. The resulting crystals were washed with water and dried.

15 [0266]

The above-obtained crystals, 100 ml of toluene and 1 g of acetic anhydride were placed in a 200-ml reaction flask provided with a magnetic stirrer, a reflux condenser and a thermometer, and refluxing with heating was conducted for 4
20 hours to give rise to a reaction. After the completion of

the reaction, the reaction mixture was cooled to room temperature. 100 ml of water was added for layer separation. The toluene layer was concentrated. The residue obtained was subjected to column chromatography to obtain 14 mg of Sym-N-

5 (2-ethylhexyl)-5-cyanotriindole. Yield: 9.2%

[0267]

FAB-Mass (NBA, Positive): $[(M+H)^+] = 757$

Melting point: 248°C

IR (KBr, cm^{-1}): 2219

10 ^1H -NMR (270 MHz, CDCl_3), δ (ppm):

0.30-0.40 (m, 9H), 0.42-1.05 (m, 33H),

1.60-1.80 (m, 3H), 4.35 (d, $J=7.2$ Hz, 6H),

7.75 (d, $J=8.2$ Hz, 3H), 7.89 (dd, $J=1$ Hz, 8.9 Hz, 3H),

7.93 (s, 3H)

15

Industrial Applicability

[0268]

The present invention provides a novel substituted Sym-triindole derivative and a process for production thereof.

20 The substituted Sym-triindole derivative is applicable to a

wide spectrum of uses, such as various electrification pre-
ventions, electrification controls, capacitors, batteries,
chemical sensors, displays, organic EL materials, solar cells,
photodiodes, phototransistors, nonlinear materials, photore-
5 fractive materials, rustproof agents, adhesives, fibers,
antistatic paints, electrodeposition paints, plating primers,
electric corrosion protections and the like.